Naphthenic Acids Category Robust Summaries

American Petroleum Institute Petroleum HPV Testing Group

Consortium # 1100997

I U C L I D Data Set

Existing Chemical: Naphthenic Acids Category

CAS No. : 1338-24-5, 64754-89-8 (and supporting chemical 61790-13-4)

EINECS Name

Producer related part

Company : Creation date :

Substance related part

Company: American Petroleum Institute

Creation date : May 15, 2012

Status

Memo : Robust summary

Printing date : Revision date : Date of last update :

Number of pages : 89

Chapter (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 Reliability (profile) : Reliability: without reliability, 1, 2, 3, 4

Flags (profile)

1. General Information

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1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type Substance type Physical status **Purity** Colour Odour

Remark : Naphthenic acid fractions are oily liquids. The salts may be liquid or solid.

Naphthenic acids (CASRN 1338-24-5, 64754-89-8, and 61790-13-4) are

classified as monobasic

carboxylic acids of the general formula RCOOH, where R represents the naphthene moiety consisting of cyclopentane and cyclohexane derivatives. Naphthenic acids are composed predominantly of alkyl-substituted cycloaliphatic carboxylic acids, with smaller amounts of acyclic aliphatic acids. The cycloaliphatic acids include single and fused multiple cyclopentane and cyclohexane rings. The carboxyl group is usually attached to a side chain rather than directly to the ring. Aromatic, olefinic, hydroxy and dibasic acids are present as minor components.

Naphthenic acids recovered from refinery streams occur naturally in the crude oil and are not formed during the refining process. Heavy crudes have the highest acid content, and paraffinic crudes usually have low acid content. Naphthenic acids are obtained by caustic extraction

of petroleum distillates, primarily kerosene and diesel

fractions.

Reference

1.1.2 SPECTRA

SYNONYMS AND TRADENAMES

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1. General Information

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1.9.2 COMPONENTS			
1.10 SOURCE OF EXPOSURE			
1.11 ADDITIONAL REMARKS			
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2. Physico-Chemical Data

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2.1 MELTING POINT

Sublimation : Method : Year :

GLP : no data

Test substance: other TS: Naphthenic acids, commercial mixtures

Remark: Values cited represent ranges of melting points cited in product literature

data and Material Safety Data Sheet for commercial naphthenic acid

products.

Result : -35 °C to + 0 °C (Soc Tech, 2003)

-35 °C to + 2 °C (AGS Chemicals, 2003)

+30 °C (Mallincrodt Baker, 1997)

Reliability : (4) not assignable

Original source data were not available for review.

Reference (2) (23) (34)

2.2 BOILING POINT

Decomposition : Method : Year : GLP :

Test substance : other TS: Naphthenic Acids (CAS Nos. 001338-24-5; 061790-13-4;

064754-89-8)

Remark: Values reported vary widely due to varied composition of the hydrocarbon

mixture in naphthenic acids. Values given represent various commercial

preparations of naphthenic acids.

Result : 250 °C to 350 °C (Soc. Tech., 20031)

140 °C to 200 °C (AGS Chemicals, 20032)

200 °C to 370 °C (Brient et al., 1995)

Reliability : (4) not assignable

Reference (3) (5) (35)

2.3 DENSITY

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Decomposition

Method : other (calculated): EPIWIN, MPBPWIN V1.40 (US EPA 2000)

Year :

GLP

Test substance: other TS: Naphthenic Acids (CAS Nos. 001338-24-5; 061790-13-4;

064754-89-8)

Remark : A search for pressure values of naphthenic acids failed to

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2. Physico-Chemical Data

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uncover reliable information. Product literature data provided narrative phrases such as "very low" or "not applicable" when describing the vapor pressure characteristic for commercial products (SocTech, S.A., 2003; AGS Chemicals Limited. 2003). To gain an understanding of vapor pressure characteristics of naphthenic acids, various naphthenic acid structures described by Brient et al. (1995) were estimated for vapor pressure using the EPIWIN computer model (U.S. EPA 2000).

The vapor pressure of complex mixtures is equal to the sum of the vapor pressures of the individual constituents in their pure form times their mole fraction in the mixture (Raoult's Law). Therefore, the total vapor pressure of a complex mixture of naphthenic acids will depend on the proportion of different molecular weight constituents making up the mixture. It is estimated from vapor pressure modeling that commercial products will have vapor pressure values near or below the measurable limits cited in standard reference guidelines (OECD Guideline 104, Vapor Pressure; OECD, 1995). Hence, based on Raoult's Law, the total vapor pressure of naphthenic acids is expected to be exceedingly low.

Result : C Mole. Vapor

Naphthenic Acid Z-No. No. Wt. Pressure, Pa

decanoic acid

0 10 172 0.049

dodecanoic acid

0 12 200 0.0021

2-methyl, 1-cyclopentyl propanoic acid

-2 10 170 0.32

4-methyl, 1-cyclohexyl decabutanoic acid

-2 21 325 0.000020

3-methyl, bicyclooctyl-[3.3]-7-propanoic acid

-4 12 196 0.042

3-methyl, bicyclodecyl-[4.4]-8-decanoic acid

-4 21 323 0.000019

3-methyl, tricyclodecapropyl-[3.3.3]-11-propanoic acid

-6 17 264 0.00056

3-methyl, tricyclodecapropyl-[3.3.3]-11-Heptanoic acid

-6 21 321 0.000019

3-methyl, tetracyclodecaheptyl-[4.2.2.2]-11 propanoic

acid -8 21 319 0.000021

Test condition : Not applicable, vapor pressures were calculated by MPBPWIN,

V1.40, EPIWIN V3.10

Reliability : (2) valid with restrictions

Estimated vapor pressures were obtained from a validated

computer program.

Reference (1) (25) (29) (33) (38)

2.5 PARTITION COEFFICIENT

Method : other (calculated): EPIWIN, KOWWIN V1.66 (US EPA 2000)

Year : 2000

GLP

Test substance: other TS: Naphthenic Acids (CAS Nos. 001338-24-5; 061790-13-4;

064754-89-8)

Remark: No partition coefficient measurements were found for

naphthenic acids. Therefore, partition coefficients for a range of molecular weight naphthenic acids were estimated

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using the EPIWIN computer model (U.S. EPA 2000). The partition coefficients reported here span the molecular weights and numbers of cycloalkane rings reported to exist in Athabasca oil sands extracts and commercial products

(Brient et al. 1995). It may be

expected, however, that the lowest molecular weight structures will have the lowest partition coefficients of

the compounds in the complex mixtures.

Result : C Mole. Log

Naphthenic Acid Z-No. No. Wt. Kow

decanoic acid

0 10 170 4.1

dodecanoic acid

0 12 200 4.6

2-methyl, 1-cyclopentyl propanoic acid

-2 10 170 3.8

4-methyl, 1-cyclohexyl decabutanoic acid

-2 21 325 9.2

3-methyl, bicyclooctyl-[3.3]-7-propanoic acid

-4 12 196 3.8

3-methyl, bicyclodecyl-[4.4]-8-decanoic acid -4 21 323 8.2

3-methyl, tricyclodecapropyl-[3.3.3]-11-propanoic acid

-6 17 264 6.0

3-methyl, tricyclodecapropyl-[3.3.3]-11-Heptanoic acid

-6 21 321 8.0

3-methyl, tetracyclodecaheptyl-[4.2.2.2]-11 propanoic

acid -8 21 319 6.3

Test condition : Not applicable, partition coefficients were calculated by KOWWIN,

V1.66, EPIWIN V3.10

Reliability : (2) valid with restrictions

Estimated partition coefficients were obtained from a validated

computer program.

Reference (5) (38)

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

PHYS CHEM WATER SOLUBILITY								
Category Chemical :	Naphthenic acids, CAS no. 1338-24	-5						
Test Substance :	Naphthenic acids, CAS no. 1338-24	-5						
Test Substance Purity/Composition and Other Test Substance Comments:	Specific analyses of the test substant Acid number: Unsaponifiables (total): Viscosity @40°C: Specific gravity @20°C: Color (Garner), GI Water content: Phenolic content (acid): Total sulfur: CP - Flash point °F (COC):	235 mg KOH/gm 4.9% 32 cst 0.969 4.5 0.07% 0.31% 0.34						
Category Chemical Result Type :	Measured							
Test Substance Result Type :	Measured							
RESULTS								
Water Solubility Indicator :								

2. Physico-Chemical Data

Water Solubility Input type:	Value or Range?		
Water Solubility Value/Range: So	plubility: = 88.1 mg/L @ Temperature: approximately 20°C		
pH Value :	Value or Lower Range: 7.5 Upper Range :		
pKa - Protein Kinase:			
pH Value at Saturation:			
Results Remarks :	The solubility value represented the measured concentration of total dissolved naphthenic acids in the water accommodated fraction of freshwater algal nutrient medium (pH 7.5) using a loading rate of 100 mg/L. Higher solubility concentrations may be achieved using higher loading rates.		
STUDY/METHOD			
Key Study Sponsor Indicator :	Key		
Year Study Performed :	2009		
Method/Guideline Followed :	Other, similar to OECD 105 flask method		
Method/Guideline and Test Condition Remarks:	A 100 mg/L loading rate solution of naphthenic acids in freshwater algal nutrient medium was prepared in an aspirator bottle containing a Teflon stir bar. Triplicate bottles were prepared in this manner. The bottles were placed on magnetic stir plates and stirred at a rate to maintain a vortex of approximately 30-50% of the static solution depth. One of the aspirator bottles was removed from the stir pates at 18, 24, and 72 hours and allowed to settle for one hour. After settling, solutions were drained from the bottom outlet of the aspirator bottle into a sample bottle. The first 100 mL was sent to waste and care was taken to ensure that no insoluble fraction was carried over into the sample bottle. Test solutions were analyzed for the concentration of naphthenic acid, using Fourier transform infrared spectroscopy (FTIR). Analysis was accomplished based on a method developed at ABC Laboratories following Jivraj et al. 1991.		
GLP:	Yes		
Study Reference :	ABC Laboratories Inc. 2009. Validation of test solution preparations and analytical methods for use in the determination of naphthenic acids in various media used in environmental toxicity studies. ABC study no. 64403, Analytical Bio-Chemistry Laboratories, Columbia, Missouri.		
RELIABILITY/DATA QUALIT	Y		
Reliability:	1 (reliable without restrictions)		
Reliability Remarks :	comparable to a guideline study		

PHYS CHEM WATER SOLUBILITY	
Category Chemical :	1338-24-5
Test Substance :	1338-24-5
Test Substance Purity/Composition and Other Test Substance Comments:	
Category Chemical Result Type:	

2. Physico-Chemical Data

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Test Substance Result Type :	
RESULTS	
Water Solubility Indicator :	
Water Solubility Input type:	Value or Range? RANGE
Water Solubility Value/Range: So	olubility: 70 mg/L to 5040 mg/L @ Temperature: 25°C
pH Value :	Value or Lower Range: 0.91 Upper Range: 9.16
pKa - Protein Kinase:	
pH Value at Saturation :	
Results Remarks :	The solubility values were provided in a general background report on naphthenic acids. The report cited the solubility data orginated from a commercial standard liquid formulation obtained from Baker Chemical Co. The original data were taken by CEATAG (1998) from Kharrat (1996).
STUDY/METHOD	
Key Study Sponsor Indicator :	
Year Study Performed :	
Method/Guideline Followed :	
Method/Guideline and Test Condition Remarks:	
GLP:	
Study Reference :	CEATAG (CONRAD Environmental Aquatics Technical Advisory Group). 1998. Naphthenic acids background information discussion report. Alberta Department of Energy, Edmonton, Alberta, Canada. 65 pp. Kharrat, A. 1996. Physico-chemical properties of naphthenic acids. Alberta Environmental Centre Progress Report October 1, 1005 – March 31, 1996. XD952287.RPT/6/4/96/PS.
RELIABILITY/DATA QUALIT	Υ
Reliability :	4 (not assignable)
Reliability Remarks :	Data retrieved from a secondary reference. The original report that contained details of the methods and results was not available for review.

Memo : Water solubility of naphthenic acids

Remark : Values of water solubility reported in product literature

data have varied widely. CEATAG (1998) reported water solubility values of one commercial product to range from 70 mg/l at pH 0.91 to 5040 mg/l at pH 9.16. Other product data sources for water solubility report narrative phrases such as "very low water solubility" (SocTech S.A., 2003), "not applicable" (Mallinckrodt Baker Inc., 1997), or "only slightly soluble in water" (AGS Chemicals Limited, 2003).

: (4) not assignable

Reliability

Data were obtained from secondary literature sources.

04.01.2005 (1) (8) (22) (33)

2. Physico-Chemical Data **Id** Naphthenic Acids **Date** May 15, 2012 2.6.2 SURFACE TENSION 2.7 FLASH POINT 2.8 AUTO FLAMMABILITY 2.9 FLAMMABILITY 2.10 EXPLOSIVE PROPERTIES 2.11 OXIDIZING PROPERTIES 2.12 DISSOCIATION CONSTANT 2.13 VISCOSITY 2.14 ADDITIONAL REMARKS

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3.1.1 PHOTODEGRADATION

Deg. product :

Method : other (calculated): EPIWIN V3.10; subroutine AOPWIN V1.90

Year :

Test substance: other TS: Naphthenic Acids (CAS Nos. 001338-24-5; 061790-13-4;

064754-89-8)

Remark : AOPWIN V1.90 calculates atmospheric oxidation rate constants

between photochemically produced hydroxyl radicals and organic chemicals. These rate constants are then used to calculate half lives for those compounds based on average atmospheric concentrations of hydroxyl radicals and ozone. Atmospheric oxidation rates were calculated for a range of molecular structures covering a range of molecular weights and ring structures that were reported to exist in Athabasca oil sands extracts and commercial products (Rogers et al.,

2002; Brient et al. 1995).

Although the low vapor pressures of these base oils indicate

that volatilization will not be a very significant fate process, oxidation half-lives indicate that any vapors

emitted to the troposphere would be rapidly oxidized and not

persist in the atmosphere.

Result : C Mole. Half

Naphthenic Acid Z-No. No. Wt. Life, days

2-methyl, 1-cyclopentyl propanoic acid

-2 10 170 0.9

4-methyl, 1-cyclohexyl decabutanoic acid

-2 21 325 0.3

3-methyl, bicyclooctyl-[3.3]-7-propanoic acid

-4 12 196 0.8

3-methyl, bicyclodecyl-[4.4]-8-decanoic acid

-4 21 323 0.3

 $\hbox{3-methyl, tricyclodecapropyl-} \hbox{[3.3.3]-11-propanoic acid}\\$

-6 17 264 0.3

3-methyl, tricyclodecapropyl-[3.3.3]-11-Heptanoic acid

-6 21 321 0.3

3-methyl, tetracyclodecaheptyl-[4.2.2.2]-11 propanoic

acid -8 21 319 0.3

Test condition: Not applicable, photodegradation potential was calculated by

AOPWIN, V1.90, EPIWIN V3.10

Reliability : (2) valid with restrictions

Estimated water solubility values were obtained from a

validated computer program.

Reference (5) (30) (38)

3.1.2 STABILITY IN WATER

Remark: Hydrolysis of an organic chemical is the transformation

process in which a water molecule or hydroxide ion reacts to form a new carbon-oxygen bond. Chemicals that have a potential to hydrolyze include alkyl halides, amides,

carbamates, carboxylic acid esters and lactones, epoxides, phosphate esters, and sulfonic acid esters (Harris, 1982).

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The chemical components found in the materials that comprise the gas oil category are hydrocarbons that are not subject to hydrolysis because they lack functional groups that hydrolyze.

Reference (15)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : Media :

Air : % (Fugacity Model Level I)
Water : % (Fugacity Model Level I)
Soil : % (Fugacity Model Level I)

Biota : % (Fugacity Model Level II/III)
Soil : % (Fugacity Model Level II/III)

Method : other: Level 1 Fugacity-Based Environmental Equilibrium Partitioning

Model (Version 2.11)

Year :

Remark : Multimedia distribution was calculated for a range of

naphthenic acids covering molecular weight and ring structures of such constituents found in Athabasca oil sands

extracts and commercial products (Rogers et al., 2002;

Brient et al., 1995).

Result : Air / Water / Soil / Sediment / Suspended Sediment / Biota

Naphthenic Acid Type (Z-number)(C-number)(Molecular Weight)

Susp

Air Water Soil Sed Sed Biota 2-methyl,1-cyclopentyl propanoic acid (-2)(10)(170)

2 16 81 1.8 <0.1 <0.1

4-methyl,1-cyclohexyl decabutanoic acid (-2)(21)(325)

<0.1 <0.1 98 2 <0.1 <0.1

3-methyl, bicyclooctyl-[3.3]-7-propanoic acid (-4)(12)(196)

0.4 15 83 2 <0.1 <0.1

3-methyl, bicyclodecyl-[4.4]-8-decanoic acid (-4)(21)(323)

<0.1 <0.1 98 2 <0.1 <0.1

3-methyl, tricyclodecapropyl-[3.3.3]-11- propanoic acid

(-6)(17)(264)

<0.1 0.1 98 2 <0.1 <0.1

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3-methyl, tricyclodecapropyl-[3.3.3]- 11 heptanoic acid

(-6)(21)(321)

<0.1 <0.1 98 2 <0.1 <0.1

3-methyl, tetracyclodecaheptyl-[4.2.2.2]-11 propanoic acid

(-8)(21)(319)

<0.1 <0.1 98 2 <0.1 <0.1

Test condition : The EQC Level I is a steady state, equilibrium model that

utilizes the input of basic chemical properties including molecular weight, vapor pressure, and water solubility to calculate distribution within a standardized regional

environment.

Reliability : (2) valid with restrictions

Estimated environmental distribution was obtained from a

validated computer program.

Reference (5) (30) (21)

3.3.2 DISTRIBUTION

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Remark: No standardized testing for ready or inherent biodegradation

was found for naphthenic acids. Results of relevant scientific journal articles on the biodegradability of naphthenic acids are reviewed in Section 3.8

Reference

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

3.8 ADDITIONAL REMARKS

Memo : Biodegradation of cycloalkane carboxylic acids in oil sand tailings

Remark : Herman et al. (1994) investigated the ability of microbial

populations indigenous to oil sands tailings to biodegrade solutions of natural naphthenic acids from oil sands

tailings and commercial naphthenic acid sodium salts (Kodak

Chemicals).

Four experiments were run:

1) Evaluation of mineralization of naphthenic acids sodium salts (NAS) and oil sands tailings extracts of naphthenic acids (TEX),

2) Evaluation of mineralization of four model naphthenic

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acid compounds, cyclohexane carboxylic acid (CCA), cyclohexane pentanoic acid (CPA) 2-methyl-1-cyclohexane carboxylic acid (2MCCA), and trans-4-pentylcyclohexane carboxylic acid (4PCCA),

- 3) Gas chromatographic analysis of NAS and TEX biodegradation, and
- 4) Respirometry measurements of cyclohexane pentanoic acid, NAS, and TEX in tailings microcosms.

Test Substances: Test substances used in the four experiments included the following materials: 1) Tailings water extract (TEX), 2) commercial sodium naphthenate mixture (NAS), and 3) pure compound naphthenic acids, cyclohexane carboxylic acid (CCA), cyclohexane pentanoic acid (CPA), 2-methyl-1-cyclohexane carboxylic acid (2MCCA), and trans-4-pentylcyclohexane carboxylic acid (4PCCA).

Inoculum: Inoculum used in the biodegradation experiments was NAS- and TEX- degrading enrichment cultures derived from oil sands tailings water. These cultures were created by diluting a 10-ml sample of oil sands tailing into 90 ml of mineral salts medium that contained either NAS (100 mg/l) or TEX (1:50 dilution). The mineral salts medium was modified Bushnell-Haas medium. Successive transfers 1% v/v) of the enrichment culture into fresh NAS- to TEX-containing medium were on monthly basis and incubated at room temperature on a gyratory shaker (100 rpm). The viable cell number within each enrichment culture was estimated using the plate count technique.

Experiment No. 1. A measurement of CO2 production was used to evaluate the ability of the enrichment cultures to mineralize components within both the NAS and TEX mixtures. Mineralization experiments were performed using 60-ml serum bottles containing 15 ml of growth medium. The growth medium consisted of sterilized mineral salts medium with NAS (100 mg/l) or TEX (1:20 and 1:50 dilutions) as the sole carbon source. Dissolve organic carbon analyses showed that 100 mg/l of NAS contained 60 mg C/l, while 1:20 and 1:50 dilutions of TEX contained 50 and 21 mg C/l, respectively. The serum bottles were inoculated with 0.15 ml of either the NAS-degrading or the TEX-degrading enrichment culture, sealed with rubber stoppers, and incubated at room temperature on a gyratory shaker (100 rpm). At 3 to 6-day intervals over 24 to 30 days, three inoculated bottles and one control (inoculated but lacking NAS or TEX) were acidified to pH <2 using 1 ml of 1M H2SO4 to convert all forms of inorganic carbon into CO2. A 0.5 ml headspace sample from each bottle was analyzed for CO2 content by gas chromatography. Mineralization of the organic substrate was first corrected for the amount of CO2 in the control bottles, then expressed either as the total amount of CO2 produced within the bottle or as the percentage of organic carbon converted to CO2.

Results of Experiment No. 1. The mineralization studies showed that the NAS- and TEX-degrading enrichment culture was capable of mineralizing components within both the NAS and TEX mixtures. The percentage of organic carbon converted to CO2 by the NAS-degrading culture was 48% (day 24) in the NAS bottles and 20% (day 20) in the TEX bottles. The percentage of organic carbon converted to CO2 by the

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TEX-degrading culture was 34% (day 30) for the TEX bottles and 20% (day 25) for the NAS bottles.

Experiment No. 2. Mineralization of the four model naphthenic acid compounds was measured as the amount of CO2 evolved from incubating solutions of the compounds dissolved in nutrient medium and inoculated with enrichment cultures of NAS-degrading microorganisms. TEX-degraders, or oil sands tailings pond water (TPW). Fifteen milliliters of 1 mM solutions of the compounds dissolved in mineral salts medium were placed in 60-ml serum bottles and inoculated (1% v/v) with the different sources of microbes then sealed with robber stoppers. Bottles were incubated at room temperature on a gyratory shaker (100 rpm). After 3, 6, 12, and 24 days, duplicate bottles were acidified and headspace CO2 determined by GC. The level of CO2 production was corrected for the amount of CO2 within the control bottles and expressed as the percentage of organic substrate converted to CO2.

Results of Experiment No. 2. The following results were obtained:.

Mineralization by day 24, % organic C converted to CO2:

Substrate	NAS-degraders TEX-degraders TPW						
CCA	41	56		57			
CPA	45	57		58			
2MCCA		47	7		67		
4PCCA		6	24		24		

Experiment No. 3. A 1% (v/v) inoculum of the NAS-degrading enrichment culture was placed in 125-ml Erlenmeyer flasks containing 50 ml of either NAS (30 mg/l) or TEX (1:50 dilution) in mineral salts medium. Control flasks received inoculum of heat-killed cells. The flasks were incubated at room temperature on a gyratory shaker (100 rpm). After an incubation period of 4, 8, and 16 days for NAS and 6, 12, and 24 days for TEX, the contents of two flasks and two control flasks were extracted for GC analysis. Samples were extracted and the carboxylic acids were derivatized to methyl esters prior to analysis. Derivatized extracts were analyzed by GC with a capillary column and flame ionization detector.

Results of Experiment No. 3. Chromatographic analysis of solution from the control flasks revealed an unresolved series of many overlapping peaks that created a hump in the GC profile. When the mixture that was inoculated with NAS-enrichment culture, a reduction in the size of the hump was evident within 4 days, indicating that components within the naphthenic acid mixture were being degraded. Chromatographic analysis of the TEX samples revealed a similar hump of many overlapping peaks that appeared in the NAS GC profile. Biodegradation of TEX by the NAS-degrading culture did not result in a noticeable reduction in the size of the hump associated with TEX, despite evidence of mineralization of components within the mixture.

Experiment No. 4. A measurement of CO2 production and O2 utilization within sealed microcosms was used to monitor microbial activity in samples of TPW, and to determine the

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effect of nutrient addition (N and P) or carbon substrate addition (cyclohexane pentanoic acid (CPA), sodium salts of naphthenic acids (NAS), or tailings pond extracts of carboxylic acids (TEX)) on the level of microbial activity within TPW.

60 ml of TPW was placed into sterile 125-ml Erlenmeyer flasks, sealed with rubber stoppers in which a sampling port had been drilled and then sealed with clear silicone. Nutrients in the form of N and P were added. Carbon substrates (CPA, NAS or TEX) were added as a filter-sterilized solution to crate a final concentration of 60 mg organic carbon/l. All flasks were incubated at room temperature on a gyratory shaker (100 rpm). At 3 to 80day intervals, 0.5 ml of headspace was sampled and analyzed for CO2 and O2 using GC. Following 5 weeks of incubation, the contents of the flasks containing CPA were extracted and analyzed using the procedure described for the GC analysis in experiment 3.

Results of Experiment No. 4. The addition of CPA to TPW resulted in increased microbial activity, as indicated by greater levels of CO2 production and O2 utilization when compared with TPW alone. Sterilized TPW demonstrated no CO2 production or O2 utilization. Even greater levels of microbial activity were evident when N and P were added in addition to CPA, indicating that mineralization could be enhanced by the addition of mineral nutrients. GC analysis of CPA in TPW microcosms after 35 d of incubation revealed that the concentration of CPA was below the level of detection in 2/3 microcosms and reduced 10-fold in the third microcosm. There was no detectable CPA in the three N and P-amended microcosms.

Similarly, NAS and TEX additions to microcosms increased microbial activity in TPW, although microbial activity was enhanced by the addition of N and P. Increases in both CO2 evolution and O2 utilization were seen.

Conclusions. This investigation showed that naphthenic acids, either as a commercial preparation of sodium salt (NAS) or natural extracts from oil sands tailing water (TEX) are capable of being utilized by natural assemblages of microorganisms. Addition of nitrogen and phosphorus enhances the utilization of these substrates by the microbes.

Reliability

(2) valid with restrictions

The report was a well-documented study that meets basic scientific principles.

Reference

(19)

Memo

Biodegradation of naphthenic acids

Remark

Herman et al. (1993) conducted four experiments on the biodegradation of specific cycloalkane carboxylic acids:

Experiment No. 1. Biodegradation of four naphthenic acid compounds (cyclopentane carboxylic acid, CCP; cyclohexane carboxylic acid, CCH; 1-methyl-1-cyclohexane carboxylic acid, 1MCCH; and 2-methyl-1-cyclohexane carboxylic acid, 2MCCH) was measured in pore water from Athabasca oil sands tailings ponds. The purpose of the tailings ponds was to

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serve as a settling basin to separate solids from liquid generated during the extraction of acidic compounds from bitumen. Therefore, the tailings ponds were considered to harbor indigenous microorganisms adapted to naphthenic acids. The collected pore water was centrifuged and filtered and served as the nutrient medium. Inoculum was 0.5 ml of the original oil sands tailings sample. Duplicate flasks containing 30 ml of medium were spiked with 1-ml aliquots of stock solutions of the different naphthenic acids to achieve a final concentration of 1000 mg/l. Test flasks received the inoculum and control flasks received inoculum in which the microbes had been heat-killed. One set of duplicate flasks received a nutrient addition in the form of NH4NO3, K2HPO4, and KH2PO4 to a final concentration of 0.2 g/l of each compound. The flasks were incubated at room temperature on a rotary shaker. After 0, 3, 6, 9, 16, 26, and 40 days, a 3-ml sample was removed, centrifuged, and filtered through a 0.2 micron syringe filter. The samples were analyzed for the test compounds by gas chromatography equipped with a flame ionization detector. Peak areas were converted to concentration using a calibration curve for each compound.

Results of Experiment 1. The bacterial populations of oil sands tailings was shown to have the metabolic capability of degrading carboxylated cycloalkanes as shown in the following table of results.

		Percent Remaining						
	C	CP	CCI	Н	MC	CH	2MC	CH
Da	y NP	- NP+	NP-	· NI	P+ N	P- N	P+ NP-	NP+
0	100	42	100	68	100	100	100	100
6	100	5	100	12	100	100	100	100
10	100	0	100	1	100	100	100	100
16	100	0	100	0	100	100	100	100
26	100	0	100	0	100	100	100	49
40	100	0	100	0	100	100	100	0

Using tailings pond water as a growth medium, degradation of CCP, CCH, and 2MCCH was achieved only if nutrients were added to the medium. CCP and CCH were degraded rapidly, within one week, while methylated carboxylic acids were more resistant to biodegradation. 2MCCH was degraded within 40 days, but no degradation was observed for 1MCCH.

Experiment No. 2. Triplicate tailings pond microcosms were created using 200 ml of the tailings sample (as inoculum and medium) in 500-ml Erlenmeyer flasks closed with cotton stoppers. A filter-sterilized solution of CCP and 1MCCH was added to each microcosm for a final concentration of 1000 mg/l. Sterile controls were autoclaved and also spiked with the test compounds. Microcosms were incubated at room temperature on a rotary shaker. After 1, 2, 3, 4, 6, and 9 weeks, samples were removed and analyzed for CCP and 1MCCH by GC.

Results of Experiment No. 2. Biodegradation of CCP was complete within the first week. No biodegradation of 1MCCH was evident after six weeks. At the six-week period, nitrogen and phosphorus was added whereby complete biodegradation of 1MCCH was noted following between the 6 and 9-week sampling. No 1MCCH was measured at 9 weeks.

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Neither CCP nor 1MCCH was degraded in the control microcosms.

Experiment No. 3: Tailings pond bacteria were isolated on agar plates and colony types were examined for their ability to utilize carboxylated cycloalkanes as their sole carbon source. Individual colonies were inoculated into a solution of carboxylated cycloalkanes (1000 mg/l) in modified Bushnell and Haas (MGH) minimal salts medium. The ability of the isolate to metabolize the carbon source was monitored by GC analysis. In a second part to this experiment, a carboxylate-degrading mixed bacterial culture was enriched from the tailings pond sample using standard procedures. The mixed culture was maintained on a mixture of CCP, 1MCCH, and 2MCCH (500 mg/l each) in MBH with yeast extract (1000 mg/l) added as a supplemental carbon source.

Results of Experiment No. 3. Of 10 separate colony types isolated from oil sands tailings, one colony type was found to utilize CCP and CCH as its sole carbon source. The isolate was a Gram negative, non-motile, catalase positive, oxidase negative, non-fermenting, aerobic rod, and was identified as an Acinetobacter sp. The isolate rapidly degraded CCP and CCH, with complete loss of substrate from the medium within 2 weeks of incubation. However, this isolate was unable to degrade methyl-substituted cyclohexane carboxylic acids. The mixed bacterial culture enriched from the tailings pond sample on a mixture of carboxylated cycloalkanes was found to degrade 1MCCH and 2MCCH, but only when the medium was supplemented with yeast extract. After a 2-week incubation period, the mixed culture had degraded 100% of the 1MCCH and 67% of the 2MCCH.

Experiment No. 4. Radiolabeled hexadecane was spiked into the maltene fraction of pure bitumen. Hexadecane mineralization experiments were performed using 5 ml of oil sands tailings in 60-ml serum vials and inoculated with 10 ul of spiked maltene. One set of vials received nutrient addition as described before. Sterile controls were autoclaved before the addition of the labeled hydrocarbon. Mineralization was determined from triplicate vials after 5, 10, 16, 27, and 40 days using the closed-loop trapping system. Radioactivity was measured using a scintillation cocktail and a Beckman LS8000 scintillation counter.

Results of Experiment No. 4. The results of hexadecane mineralization within oil sands tailings showed that the biodegradation of an n-alkane was nutrient limited. Percent biodegradation reached 50% by day 16 and maintained a plateau through day 40.

Conclusions. This study showed the potential for biodegradation of naphthenic acids by investigating the biodegradation of both carboxylated cycloalkanes and hexadecane. Although natural naphthenic acids present in oil sands tailings have greater structural complexity than the compounds examined in this study, the results show the potential for both for biodegradation of the alkyl side chain and the carboxylated cycloalkane ring components of naphthenic acids. Biodegradation potential was reduced by methyl substitution on the cycloalkane ring, although these compounds could be degraded with the addition of mineral

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nutrients.

Reliability : (2) valid with restrictions

The report was a well-documented study that meets basic

scientific principles.

Reference (18)

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4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type : static

Species: Brachydanio rerio (Fish, fresh water)

Exposure period : 96 hour(s)

Unit

Limit test : Analytical monitoring : no

Method : Year : 1965 GLP : no

Test substance : other TS: Naphthenic Acids

Method : Statistical Method:

Graphical interpolation for determining the LC50.

Result : 96-hour TLm = 16.3 ppm

The following dose-response data were provided:

Concentration of	Numb	oer	% Dead at
Naphthenic acids, p	pm Teste	ed 96 ho	urs
0 (control #1)	10	0	
0 (control #2)	10	0	
7.5	10	0	
8.7	10	40	
10	10	20	
11.5	10	0	
13.5	10	20	
15.5	10	30	
18	10	80	
21	10	100	
24	10	100	

The article reported that pH and dissolved oxygen concentrations were taken during the test, but these data were not reported.

Test condition

Test containers were 2.5 gallon aquariums, each fitted with an air stone through which compressed air was bubbled to maintain a 5-9 ppm dissolved oxygen concentration in the dilution water. The aquariums were maintained at a temperature of 24 +/- 1 C. Dilution water was synthetic soft water prepared with distilled water and ACS grade chemicals.

The lot of test fish displayed no visible disease. The average size was 3.2 cm total length. Before testing the fish were acclimated to the dilution water for 5 days. During the acclimation period they were fed Daphnia and white worms, but were not fed for 36 hours before or during the testing.

Test concentrations were prepared by direct addition of the test substance to the test chambers followed by mixing. Test concentrations were control, 7.5, 8.7, 10, 11.5, 13.5, 15.5, 18.0, 21.0, and 24.0 ppm naphthenic acids. After the test solutions were prepared, ten fish were placed in each test container. Controls were run in duplicate, while test levels were run singly. Mortality was evaluated at 24, 48, and 96 hours, and the criteria for death was a cessation of

4. Ecotoxicity Id Naphthenic Acids

Date May 15, 2012

gill movement and failure to respond to mechanical stimulus.

Following the 96 hour test period the TLm (median tolerance

limit) was determined from visual observation of the dose-response pattern. Where no exact TLm response resulted, the TLm was interpolated from a plot of the concentration and survival data on semi-log paper.

Reliability : (2) valid with restrictions

The test was conducted under referenced test conditions current for the period in which the study was run. The report provided sufficient details for assessment.

(6) (10) (16)

Type : static

Species: Gasterosteus aculeatus (Fish, estuary, marine)

Exposure period : 96 hour(s)
Unit : mg/l

Limit test :

Analytical monitoring : no Method : Year :

GLP : no data

Test substance : other TS: Naphthenic acid mixture (commercially available from Eastman

Chemicals)

Result: LC50 estimated to be in the range of 5 mg/l.

The following dose response data were reported:

Concentration (mg/l) % Survival 0 (control) 100 2.5 60 5 10

15 0 30 0

Although an LC50 could have been calculated using contemporary methods, the author elected to estimate its value. The report stated that water chemistry data were

collected but no data were reported.

Test condition: Summary of Test Conditions

Organism age: juvenile
Test Temperature: $20 \,^{\circ}\text{C} \pm 2 \,^{\circ}\text{C}$ Photoperiod: $16 \,^{\circ}\text{h}$ light/8 h dark
Light intensity: $10 - 50 \,^{\circ}\text{micro-einsteins}$

Light quality: wide spectrum fluorescent

Test container: 5 gallon aquaria
Dilution water: Carquinex Strait

Test Volume: 15 liters
Animals per container: 10
Replicate containers: 2

Number of concentrations: 6 (5 concentrations and a

control)

Food: none
Test duration: 96 h
Test endpoint: mortality
Salinity 15 parts per thousand
Test pH: ambient

Test article: Martinez Refinery effluent

(non-toxic)

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with added

naphthenic acids

Test solutions were prepared by creating a 1 percent solution using non-toxic effluent pH adjusted to 12 with sodium hydroxide. The stock solution was mixed overnight prior to use. The stock solution was used to spike non-toxic treated effluent to nominal naphthenic acid concentrations from 2.5 to 30 mg/l.

Test organisms were held at least seven days prior to testing in dilution water. During testing at 24-h intervals, the salinity, temperature, pH, and dissolved oxygen were measured in all control and test tanks. Survival data were taken at 24-h intervals and dead individuals were removed when observed.

Reliability : (2) valid with restrictions

A statistically-defined LC50 was not calculated. Water

chemistry data were not reported.

Reference (9) (26) (36)

Category Name: RECLAIM	ED SUBSTANCES - NAPH	THENIC ACIDS		
Category Chemical :	Naphthenic acids, CAS no. 1338	-24-5		
Test Substance :	Naphthenic acids, CAS no. 1338	-24-5		
	Specific analyses of the test sub-	stance:		
Test Substance Purity/Composition and Other Test Substance Comments :	Acid number: Unsaponifiables (total): Viscosity @40°C: Specific gravity @20°C: Color (Garner), GI Water content: Phenolic content (acid): Total sulfur: CP - Flash point °F (COC):	235 mg KOH/gm 4.9% 32 cst 0.969 4.5 0.07% 0.31% 0.34		
Category Chemical Result Type:	Measured			
Test Substance Result Type:	Measured			
Method				
Year Study Performed :	2010			
Method/Guideline Followed:	OECD 203			
Deviations from Method/Guideline :	There was a brief temperature ex 22±1°C.	xcursion outside the boundaries of		
Species:	Pimephales promelas (fathead minnow)			

Yes
Yes
Renewal
3.8-L glass jars
Modified well water
0 (control), 1.3, 2.5, 5.0, 10, and 20 mg naphthenic acids/L
Nominal WAF loading rates: 0 (control), 1.3, 2.5, 5.0, 10, and 20 mg naphthenic acids/L Mean measured: 0 (control), 0.90, 2.08, 3.22, 604, 13.8 mg naphthenic acids/L
96 hours

Vehicle Used:	None				
Vehicle Name:					
Vehicle Amount and Units:					
Alkalinity:	148 mg L				
Dissolved Oxygen:	7.1 to 9.3 mg/L				
pH Value:	Value or Lower Range : 8.0 Upper Range : 8.4				
Test Temperature and Units:	Value or Lower Range : Upper Range :				
Photo (Light/Dark):	16 h light / 8 h d Light intensity: 5	dark 523 lux			
Salinity:	Freshwater				
тос:					
Water Hardness:	Value or Lower Range: Upper Range:	134 mg/L			

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Exposure solutions were prepared as water accommodated fractions (WAF). Each WAF was prepared independently based on the selected loading rates used for the test. The WAFs were prepared by adding the appropriate amount of test substance to 4 L of dilution water in clean 5-L glass carboys. Each carboy contained a 2-inch Teflon-coated stir bar and was sealed with a screw cap. The mixtures were stirred for 24 hours at a speed that created a vortex of 30-50% of the solution depth. After the stirring period, the solutions were permitted to settle for approximately 1 hour. WAFs were siphoned from the bottom of the mixing vessels, with the first ~100 mL being discarded. This procedure was repeated three times in order to prepare renewal solutions for the 24, 48, and 72-h time points of the test.

Method/Guideline Test Conditions Remarks:

The different exposure levels were established in single 3.8-L glass jars holding approximately 2.0 L of solution. At the beginning of the test, fish were impartially added one at a time to each test vessel until each vessel held its complement of 7 fish. During renewal periods, fish were transferred to freshly-prepared exposure solutions. Observations for mortality, moribundity, and sublethal responses were made every 24 hours (±1 hour).

Measurements of the concentrations of dissolved naphthenic acids in the WAFs were made on samples taken at 0 hours (fresh), 24 hours (old), 72 hours (fresh), and 96 hours (old). The method of analysis included aqueous sample extraction by methylene chloride with detection by Fourier transformed Infrared spectroscopy. The minimum quantifiable limit (MQL) for the method was 0.6 mg naphthenic acids/L. Additional characterization of the exposure solutions included analysis by gas chromatography-mass spectroscopy. This method allowed the proportion of dissolved naphthenic acids to be resolved into families of naphthenic acids having similar carbon numbers and ring numbers.

Limit Test:

No

Test Results

NOEC/LOEC/NOELR/LOELR

	Exposure Duration:	Exposure Units:	Value Description:	Value or Lower Range:	Upper Range:	Units:	Basis for Concentration:
NOEC:	96	Hours	=	3.22		ma/I	arithmetic mean measured
LOEC:	96	Hours	=	= 6.04 mg	mg/L	arithmetic mean measured	
NOELR:	96	Hours	=	5.0		mg/L	Nominal
LOELR:	96	Hours	=	10		mg/L	Nominal

LC/EC/IC/EL/LL Mean Value

Exposure Duration:	Exposure Units:	Туре	%:	Value Description:	Mean Value or Lower Mean Value:	Upper Mean Value:	Units:	Basis for Effect:	Basis for Concentration:			
24	Hours	LL	50	>	20		mg/L	Mortality	Nominal			
48	Hours	LL	50	=	11		mg/L	Mortality	Nominal			
72	Hours	LL	50	=	11		mg/L	Mortality	Nominal			
96	Hours	LL	50	=	9.0		mg/L	Mortality	Nominal			
24	Hours	LC	50	>	13.8		mg/L	Mortality	arithmetic mean measured			
48	Hours	LC	50	=	7.22		mg/L	Mortality	arithmetic mean measured			
72	Hours	LC	50	=	7.22		mg/L	Mortality	arithmetic mean measured			
96	Hours	LC	50	=	5.62		mg/L	Mortality	arithmetic mean measured			
Results Rem				fish found in determined Concentration solutions reconcentration concentration contained 1 naphthenic Other than	The LC/LR50 values of the test were based on the percentage of dead fish found in each treatment level. The NOEC/LR values were determined based on scientific judgment of the dose-response pattern. Concentrations of dissolved naphthenic acids in the fresh and old test solutions remained stable over the renewal periods. The measured concentrations in the old solutions were at least 89% of the initial concentrations. Analysis by GC-MS for carbon number and ring distribution indicated 82 – 90% of the dissolved naphthenic acids contained 10 to 16 carbon atoms with a prevalence of one and two ring naphthenic acid isomers. Other than a brief temperature excursion beyond the guideline requirements, this study met the method guideline acceptability criteria							
Reliability	//Data Q	ualit	y	1								
Reliability:				1								
Reliability Re	emarks:			Reliable wit	hout restrict	ions						
Key Study Sp	onsor Indi	cator	:	Key	Key							
Reference				Garke A 20	10 Acute to	vicity of w	ator acc	ommodator	I fractions of			
Reference:				naphthenic a determined u approach. Al	Gerke, A. 2010. Acute toxicity of water accommodated fractions of naphthenic acids to the fathead minnow, Pimephales promelas, determined under static-renewal test conditions using a step-down approach. ABC Study no. 64406, Analytical Bio-Chemistry Laboratories, Columbia, Missouri.							

Acute Toxicity to Aquatic Vertebrates							
Category Name: RECLAIMED SUBSTANCES - Naphthenic Acids							
Category Chemical:	61790-13-4						
Test Substance :	61790-13-4						

Test Substance Purity/Composition and Other Test Substance Comments :	Commercial naphthenic acids (sodium salt) was a 50% (w/v) aqueous solution supplied by Pfaltz-Bauer Inc.
Category Chemical Result T	ype: Estimated by supporting chemical
Test Substance Result Type	: measured
Method	
Year Study Performed :	
Method/Guideline Followed	:
Deviations from Method/Guideline :	
Species:	
GLP:	
Analytical Monitoring :	
Test Type:	
Test Vessel:	
Water Media Type:	
Test Concentrations:	
Nominal and Measured Concentrations:	
Total Exposure Period:	
Vehicle Used:	
Vehicle Name:	
Vehicle Amount	and Units:
Alkalinity:	
Dissolved Oxyg	en:
pH Value:	Value or Lower Range : Upper Range :

	Test Tempe and Units:	eratur	e		L	-	r Range : Range :						
	Photo (Light	Photo (Light/Dark):											
	Salinity:												
	тос:												
	Water Hardi	ness:			Lo	alue or wer R oper R	ange:						
	Guideline ditions Remarks	s:											
Limit Tes	t:												
Test Re	esults												
					NOEC/L	OEC/	NOELR/	LOELF	ł				
	Exposure Duration:		cpos Unit		Valu Descript	_	Value Lowe Rang	er	-	per nge:	Units:	C	Basis for Concentration:
NOEC:													
LOEC:													
NOELR:													
					LC/EC/I	C/EL/	LL Mear	ı Valu	е				
Exposu Duratio		Туре	/na 0/a ·		alue ription:	Va Lo	lean lue or ower lean alue:	Me	per ean lue:	Units:	Basis Effec		Basis for Concentration:
Results R	emarks:			Jap nap cor afte Em Pre and abo cor effe (NO cor	panese monthenation for the contraction of the cont	nedak de solu ons. F dation at sunt defe atera eart le es we ons (dentrate ere ca on for	a (Orizia utions we for both and con rvived we ormities I curvatied to sy ere prese defined tion (LO Iculated deform	as lati, ere ever specion tinue vere months of the ent. Do as the ent. Do as the ent. I for entities wength	pes) evaluations, expenses of untiles, expenses of the second the ach spwas 1.	embryosed over posure I the hared for l cluded spine. F ilatory I ity and netric m no obs becies. l 67 mg/ tch was	s expose a range to the atching body le optic-common growth nean of served of the control of the	ed to e of reatment of the legal of the lega	naphthenic acids ments began soon e was met. at hatch. ic irregularities pericardial edema nd optic-cephalic

Reliability/Data Quality	
Reliability:	
Reliability Remarks:	
Key Study Sponsor Indicator:	
Reference	
Reference:	Peters, L.E., M. MacKinnon, T. Fan Meer, M.R. van den Heuvel, and D.G. Dixon. (2007). Effects of oil sands process-affected waters and naphthenic acids on yellow perch (<i>Perca flavescens</i>) and Japanese medaka (<i>Orizias latipes</i>) embryonic development. Chemosphere 67:2177-2183.

Acute Toxicity to Aquatic Ve	ertebrates							
Category Name:								
Category Chemical:	61790-13-4							
Test Substance :	61790-13-4							
Test Substance Purity/Composition and Other Test Substance Comments:	The test substance was a dense, amber-colored mass of naphthenic acids – sodium salts (8-10% sodium) purchased from Acros Organics.							
Category Chemical Result Type :	measured							
Test Substance Result Type:	measured							
Method								
Year Study Performed :								
Method/Guideline Followed:								
Deviations from Method/Guideline :								
Species:	yellow perch (<i>Perca flavescens</i>)							
GLP:	no data							
Analytical Monitoring :	no							
Test Type:	semi-static							
Test Vessel:	no data							
Water Media Type:	freshwater							
Test Concentrations:	nominal							
Nominal and Measured Concentrations:	0 (control), 0.9, 1.8, and 3.6 mg/L							

Vehicle Used: Vehicle Name: Vehicle Amount and Units: Alkalinity: Dissolved Oxygen: 8.92 pH Value: Value or Lower Range: 8.38 Upper Range: Upper Range: 18.4 Upper Range: Photo (Light/Dark): 16/8 hours Salinity: O.3 ppt TOC: Water Hardness: Value or Lower Range: Upper Range:	Total Exp	osure Period:		21 days					
Vehicle Amount and Units: Alkalinity: Dissolved Oxygen: 8.92 pH Value: Value or Lower Range: 8.38 Upper Range: Test Temperature and Units: Upper Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Photo (Light/Dark): 16/8 hours Salinity: 0.3 ppt TOC: Water Hardness: Value or Lower Range: Upper Range: Units: Basis for Concentration: NOEC: Upper Range: Units: Basis for Concentration: NOEC: Upper Range: Units: Concentration:		Vehicle Used	:						
Alkalinity: Dissolved Oxygen: 8.92 pH Value: Value or Lower Range: 8.38 Upper Range: Test Temperature and Units: Upper Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Value or Lower Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Value or Lower Range: Upper Range: 18.4 Value or Lower Range: Upper Range: Units: Basis for Concentration: NOEC: Units: Basis for Concentration: NOEC: Upper Range: Units: Description: Upper Range:		Vehicle Name) :						
Alkalinity: Dissolved Oxygen: 8.92 pH Value: Value or Lower Range: 8.38 Upper Range: Test Temperature and Units: Upper Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Value or Lower Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Value or Lower Range: Upper Range: 18.4 Value or Lower Range: Upper Range: Units: Basis for Concentration: NOEC: Units: Basis for Concentration: NOEC: Upper Range: Units: Description: Upper Range:									
Dissolved Oxygen: 8.92 pH Value: Value or Lower Range: 8.38 Upper Range: Test Temperature and Units: Value or Lower Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Upper Range: 18.4 Upper Range: Upper Range: 18.4 Water Hardness: Value or Lower Range: Upper Range		Vehicle Amou	ınt and Uni	ts:					
PH Value: Value or Lower Range: 8.38 Upper Range: Value or Lower Range: 18.4 Upper Range: Upper Range: Value or Lower Range: Upper Range: Units: Description: Description: Units: Description: D		Alkalinity:							
Test Temperature and Units: Photo (Light/Dark): Salinity: O.3 ppt TOC: Water Hardness: Value or Lower Range: Upper Range: Units: Description: Upper Range: Units: Units: Description: Upper Range: Units: Units: Description: Upper Range: Units: U		Dissolved Ox	ygen:		8.92				
Test Temperature and Units: Photo (Light/Dark): 16/8 hours Salinity: 0.3 ppt TOC: Water Hardness: Value or Lower Range: Upper Range: Units: Basis for Concentration: NOEC: Upper Range: Units: Basis for Concentration: Upper Range: Units: Upper Range: Upper Range		pH Value:				inge : 8.38	Upper Range :		
Salinity: 0.3 ppt TOC: Water Hardness: Value or Lower Range: Upper Range: Upper Range: Limit Test: no Test Results NOEC/LOEC/NOELR/LOELR Exposure Duration: Exposure Units: Description: Value or Lower Range: Upper Range: Units: Basis for Concentration: NOEC: LOEC:		Test Tempera and Units:	nture						
Water Hardness: Value or Lower Range: Upper Range: Upper Range: Upper Range: Limit Test: no Test Results NOEC/LOEC/NOELR/LOELR Exposure Duration: Value Description: Value or Lower Range: Units: Basis for Concentration: NOEC: Units: NOEC: Units: Description: Des		Photo (Light/	Dark):	1	.6/8 hour	S			
Water Hardness: Value or Lower Range: Upper Range: Upper Range: Limit Test: no Test Results NOEC/LOEC/NOELR/LOELR Exposure Duration: Value Description: Value or Lower Range: Units: Basis for Concentration: NOEC: Upper Range: Value Or Lower Range: Units: Concentration:		Salinity:		C					
Water Hardness: Lower Range: Upper Range: Upper Range: Upper Range: Upper Range: Upper Range: Limit Test: no Test Results NOEC/LOEC/NOELR/LOELR Exposure Duration: Value or Lower Range: Units: Basis for Concentration: NOEC: Units: Units: Units: Concentration: NOEC: Units:		тос:							
Test Conditions Remarks: Limit Test: no Test Results NOEC/LOEC/NOELR/LOELR Exposure Duration: Value Description: Value or Lower Range: Units: Basis for Concentration: NOEC:		Water Hardne	ss:		Lower Ra				
Test Results NOEC/LOEC/NOELR/LOELR Exposure Duration: Exposure Units: Value or Lower Range: Units: Basis for Concentration: NOEC: LOEC: LOEC: LOEC: LOEC: LOEC LOEC LOEC LOEC LOEC LOEC LOEC LOEC									
NOEC/LOEC/NOELR/LOELR Exposure Duration: Exposure Units: Value Description: Value or Lower Range: Units: Basis for Concentration: Concentration:	Limit Test	:	no						
Exposure Duration: Duration Exposure Units: Value Description: Value or Lower Range: Units: Basis for Concentration: Units: Uni	Test Re	sults							
NOEC: LOEC: Value Units: Description: Lower Range: Units: Description: Lower Range: Units: Description: Lower Range: Units: Description: Concentration: Conc				NOEC/	LOEC/N	OELR/LOEL	.R		
LOEC:		Exposure Duration:	Exposure Units:	Va Descri	lue iption:	Lower	Upper Range:	Units:	Basis for Concentration:
	NOEC:								
NOELR:	LOEC:								
	NOELR:								

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Exposure Duration:	Exposure Units:	Туре	%:	Value Description:	Mean Value or Lower Mean Value:	Upper Mean Value:	Units:	Basis for Effect:	Basis for Concentration:			
96	hours	LC	100	=	3.6		mg/L	mortality	nominal			
				prior to init weeks in a	iating exper static-renev	iments. Gr val designe	oups of d syste	perch were m to contro	neld for two days exposed for thre I, 0.9, 1.8, and 3			
Results Remarks:				(8-10% sociacid concertransform in Following to the head body weightliver somation Slides of giver somation categories, cytoplasmic organism to Complete for treatment. It is predominated to for The predominated to for commercial epithelial, of acid concerts the social concerts the soci	dium) was on trations we infrared specially and severing and liver to the control of the control of the consequent ish exposed in another to the consequent ish exposed in aphthenic	btained from the measure of the spin weight were calculated issue were expected alternations of the control of	m Acrosed in the FTIR). erch we al cord for eartions wative, in general within 9 liver his faration ells. Howells.	e exposure se exposure expo	the 3.6 mg/L / comparisons we fish.			
Reliabilit	y/Data Q	ualit	y									
Reliability:				,	2 (reliable with restrictions)							
Reliability R	emarks:				Test concentrations were not measured and a complete description of the dose-response pattern was not provided.							
Key Study S	•	icator	r:	no								
Reference	2											
Reference:				Nero, V., A. Farwell, L.E.J. Lee, T. Van Meer, M.D. MacKinnon, and D.G. Dixon. (2006). The effects of salinity on naphthenic acid toxicity to yellow perch: gill and liver histopathology. Ecotoxicol Environ Safety 65(2):252-264.								

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Acute Toxicity to Aquatic Invertebrates

Category Name: RECLAIMED SUBSTANCES - NAPHTHENIC ACIDS

Category Chemical:	Naphthenic acids, CAS no. 1338-24-5					
Test Substance :	Naphthenic acids, CAS no. 1338-24-5					
Test Substance Purity/Composition and Other Test Substance Comments :	Unsaponifiables (total): Viscosity @40°C: Specific gravity @20°C: Color (Garner), GI Water content: Phenolic content (acid): Total sulfur:	bstance: 235 mg KOH/gm 4.9% 32 cst 0.969 4.5 0.07% 0.31% 0.34				
Category Chemical Result Type :	Measured					
Test Substance Result Type:	Measured					
Method						
Year Study Performed :	2010					
Method/Guideline Followed:	OECD 202					
Deviations from Method/Guideline :	None noted					
Species:	Daphnia magna					
GLP:	Yes					
Analytical Monitoring :	Yes					
Test Type:	Renewal					
Test Vessel:	250-mL glass jars					
Water Media Type:	Aged laboratory well water					
Test Concentrations:	0 (control), 5.0, 10, 20, 40, and concentrations expressed as load	d 80 mg naphthenic acids/L (test ading rate)				
Nominal and Measured Concentrations:	mg naphthenic acids/L	(control), 5.0, 10, 20, 40, and 80 ol), 3.90, 7.68, 17.0, 33.3, and 69.0				
Total Exposure Period:	48 hours					

Id Naphthenic Acids

Date May 15, 2012

Vehicle Used:	None					
Vehicle Name:						
Vehicle Amount and Units:						
Alkalinity:	148 mg/L					
Dissolved Oxygen:	6.8 to 8.8 mg/L					
pH Value:	Value or Lower Range : 7.5 Upper Range : 8.6					
Test Temperature and Units:	Value or Lower Range : 20.6 Upper Range : 22.0					
Photo (Light/Dark):	16 h light / 8 h dark Light intensity: 521 lux					
Salinity:	N/A (freshwater)					
тос:						
Water Hardness:	Value or Lower Range: Upper Range:	150 mg/L				

Exposure solutions were prepared as water accommodated fractions (WAF). Each WAF was prepared independently based on the selected loading rates used for the test. Each WAF was prepared by adding the appropriate amount of test substance to 4 L of dilution water in a clean 4-L glass carboy. Each carboy contained a 2-inch Teflon-coated stir bar and was sealed with parafilm. The WAF preparations were stirred for 24 ± 1 hours at a speed that created a vortex of 30-50% of the solution depth. After the stirring period, the solutions were permitted to settle for approximately 1 hour. The WAF was siphoned from the bottom of the mixing vessel, with the first ~100 mL being discarded. Enough WAF was collected to prepare four replicate test chambers per treatment. This procedure was repeated in order to prepare renewal solutions for the 24-h time point of the test.

Method/Guideline Test Conditions Remarks:

Each replicate test vessel contained 200-mL of the WAF or control solution. At the beginning of the test, five neonate daphnids (<24-h old) were added to each test vessel in a random process. At 24-hours into the test, the daphnids were transferred to fresh WAF solutions. Observations for immobile daphnids and sub-lethal responses were made every 24 hours (±1 hour).

Measurements of the concentrations of dissolved naphthenic acids in the WAFs were made on samples taken at 0 hours (fresh), 24 hours (fresh and old), and 48 hours (old). The method of analysis included aqueous sample extraction by methylene chloride with detection by Fourier transformed Infrared spectroscopy. The minimum quantifiable limit (MQL) for the method was 0.6 mg naphthenic acids/L. Additional characterization of the exposure solutions included analysis by gas chromatography-mass spectroscopy. This method allowed the proportion of dissolved naphthenic acids to be resolved into families of naphthenic acids having similar carbon numbers and ring numbers.

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Limit Test:	No

Test Results

NOEC/LOEC/NOELR/LOELR

	Exposure Duration:	Exposure Units:	Value Description:	Value or Lower Range:	Upper Range:	Units:	Basis for Concentration:
NOEC:	48	Hour	=	7.68		ma/I	arithmetic mean measured
LOEC:	48	Hour	=	17.0		ma/I	arithmetic mean measured
NOELR:	48	Hour	=	10		mg/L	nominal
LOELR:	48	Hour	=	20		mg/L	nominal

LC/EC/IC/EL/LL Mean Value

Exposure Duration:	Exposure Units:	Туре	%:	Value Description:	Mean Value or Lower Mean Value:	Upper Mean Value:	Units:	Basis for Effect:	Basis for Concentration:
24	Hours	EC	50	=	23.8		mg/L	immobile	arithmetic mean measured
48	Hours	EC	50	=	20.0		mg/L	Immobile	arithmetic mean measured
24	Hours	EL	50	=	28.3		mg/L	Immobile	nominal
48	Hours	EL	50	=	24.0		mg/L	immobile	nominal

The control and test solutions were clear and colorless with no visible signs of un-dissolved test substance, precipitate, or surface film throughout the study.

The endpoints of the test were based on the percentage of immobile daphnids found in each treatment level. The observation of "floating daphnids" was observed at 24 and 48 hours in the solutions prepared at the 20 mg/L loading rate. This effect was considered a sublethal effect by the testing laboratory and was used to define the NOEC(LR)/LOEC(LR), but was not included in the calculation of the study endpoints.

Results Remarks:

Concentrations of dissolved naphthenic acids in the fresh and old test solutions remained stable over the renewal period. The measured concentrations in the old solutions were at least 87% of the initial concentrations. Analysis by GC-MS for carbon number and ring distribution indicated 85-91% of the dissolved naphthenic acids contained 10 to 16 carbon atoms with a prevalence of one and two ring naphthenic acid isomers.

This study met all guideline requirements of acceptability criteria.

Reliability/Data Quality

Reliability:	1
Reliability Remarks:	Reliable without restriction
Key Study Sponsor Indicator:	Key

Reference

Reference: Rebstock, M. 2010. Acute toxicity of water accommodated fractions of

Id Naphthenic AcidsDate May 15, 2012

naphthenic acids to the water flea, Daphnia magna, determined under static-renewal conditions. ABC study no. 64404, Analytical Bio-Chemistry Laboratories, Columbia, Missouri.

Acute Toxicity to Aquatic Invertebrates					
Category Name: RECLAIMED	SUBSTANCES: Naphthenic acids				
Category Chemical :	1338-24-5				
Test Substance :	1338-24-5				
Test Substance Purity/Composition and Other Test Substance Comments :	calcium naphthenate				
Category Chemical Result Type :	estimated by supporting chemical				
Test Substance Result Type:	Measured				
Method					
Year Study Performed :					
Method/Guideline Followed:					
Deviations from Method/Guideline	:				
Species:	Nitocra spinipes				
GLP:	no data				
Analytical Monitoring :					
Test Type:					
Test Vessel:					
Water Media Type:	brackish water				
Test Concentrations:					
Nominal and Measured Concentrations:					
Total Exposure Period:	96 hours				
Vehicle Used:					
Vehicle Name:					

	Vehicle Amo	unt a	nd l	Jnits:									
	Alkalinity:												
	Dissolved Oxygen:												
	pH Value:					Value or Lower Range : Upper Range :							
	Test Temperature and Units:				Value o Lower F Upper F	Range	:						
	Photo (Light,	Photo (Light/Dark):											
	Salinity:				7 parts	per th	nousan	d					
	тос:												
	Water Hardno	ess:			Value or Lower R Upper R	ange:							
	Guideline Iditions Remar	ks:			Bengtsso chemicals	n, O. S s and p (<i>Albur</i>	Svanber pesticide <i>nus alb</i> e	g, and e formu urnus)	G. Su lation	ndstrom is agains	. 1983. T t two bra	shed by Lind The acute tox ackish water oppepod (<i>Nitro</i>	icity of 78 organisms,
imit Te	st:				No								
Test Re	sults				'								
					NOEC/L	OEC/I	NOELR/	'LOELI	R				
	Exposure Duration:		xpos Unit		Value Description:		Value or Lower Range:		Upper Range:		Units:	Units: Basis for Concentration	
NOEC:													
LOEC:													
NOELR:													
LOELR:													
					LC/EC/I	C/EL/	LL Mea	n Valu	ıe				
Exposu Duratio		Туре	% :	_	Value Description:		ean ue or ower ean alue:	Up _l Me Val	an	Units:	Basis f		sis for entration:
96	hours	LC	50	=		4.8					mortality	y Nominal	
esults I	Remarks:	1	napl for t	nthenio he nap	c acids, C	CAS 1: e ion	338-24 of the o	-5. Th	ie EC n salt	HA doss	sier cond l be valid	ration dossi cludes that t d for naphth	the LC50

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Reliability/Data Quality							
Reliability:	2 (reliable with restrictions)						
Reliability Remarks: The endpoint was determined for a supporting substance (structural surrogate of the test substance).							
Key Study Sponsor Indicator:	no						
Reference							
Reference:	Linden, E., B.E. Bengtsson, O. Svanberg, and G. Sundstrom. 1983. The acute toxicity of 78 chemicals and pesticide formulations against two brackish water organisms, the beak (<i>Alburnus alburnus</i>) and the harpacticoid copepod (<i>Nitroca spinipes</i>). Chemosphere 8:843-851.						

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Acute Toxicity to Aquatic Plants							
Category Name	RECLAIMED SUBSTANCES - NAPHTHENIC ACIDS						
Category Chemical:	Naphthenic acids, CAS no. 1338-24-5						
Test Substance :	Naphthenic acids, CAS no. 1338-24-5						
Test Substance Purity/Composition and Other Test Substance Comments :	Specific analyses of the test substance: Acid number: Unsaponifiables (total): Viscosity @40°C: Specific gravity @20°C: Color (Garner), GI Water content: Phenolic content (acid): Total sulfur: CP - Flash point °F (COC): 235 mg KOH/gm 4.9% 4.9% 4.9% 4.9% 4.9% 4.5 0.969 4.5 0.07% 9.07% 9.07% 9.034 0.34 0.34						
Category Chemical Result Type :	Measured						
Test Substance Result Type:	Measured						
Method							
Year Study Performed :	2010						
Method/Guideline Followed:	OECD 201 and OPPTS 850.5400						
Deviations from Method/Guideline:	None noted						
Species:	Pseudokirchneriella subcapitata						
GLP:	Yes						

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Analytical Monitoring :	Yes
Test Type:	Static
Test Vessel:	250-mL Erlenmeyer flasks
Water Media Type:	Algal nutrient medium prepared to ASTM E1217-97a recipe
Test Concentrations:	0 (control), 2.5, 5.0, 10, 20, and 80 mg naphthenic acids/L
	Nominal WAF loading rates: 0 (control), 2.5, 5.0, 10, 20, and 80 mg naphthenic acids/L
Nominal and Measured Concentrations:	72-h Mean measured: <mql (control),="" 1.69,="" 15.0,="" 28.9,="" 3.48,="" 44.9="" 7.38,="" acids="" and="" l<="" mg="" naphthenic="" th=""></mql>
	96-h Mean measured: <mql (control),="" 1.64,="" 14.8,="" 28.4,="" 3.51,="" 44.8="" 7.41,="" acids="" and="" l<="" mg="" naphthenic="" th=""></mql>
Total Exposure Period:	96 hours

Vehicle Used:	None	
Vehicle Name:		
Vehicle Amount and Units:		
Alkalinity:		
Dissolved Oxygen:		
pH Value:	Value or Lower Range: 6.8 Upper Range: 8.9	
Test Temperature and Units:	Value or Lower Range : 23.2 °C Upper Range : 24.2 °C	
Photo (Light/Dark):	Continuous lighting Intensity: 4357 to 4527 lux	
Salinity:		
тос:		
Water Hardness:	Value or Lower Range: Upper Range:	

Method/Guideline Test Conditions Remarks: Exposure solutions were prepared as water accommodated fractions (WAF). Each WAF was prepared independently based on the selected loading rates used for the test. Each WAF was prepared by adding the appropriate amount of test substance to 4 L of nutrient medium in a clean, autoclaved 4-L glass carboy. Each carboy contained a 2-inch Teflon-coated stir bar and was sealed with a screw cap. The WAF preparations were stirred for 24 ± 1 hours at a speed that created a vortex of 30-50% of the solution depth. After the stirring period, the solutions were permitted to settle for approximately 1 hour. The WAF was siphoned from the bottom of the mixing vessel, with the first ~ 100 mL being discarded. Enough WAF was created to prepare 7 replicate test vessels per treatment. Replicates designated A, B, C, D, E,

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and F were 250-mL Erlenmeyer flasks filled with 100 mL of WAF or control solution. Replicate G was a 1-L Erlenmeyer flask containing 600 mL of WAF or control solution. This replicate was placed beside the other replicates during the test, but served only to provide sufficient volume of solutions for analytical measurements at 72 hours. At 96 hours, replicates D, E, and F were pooled to provide the volume needed for analytical measurements at the end of the test.

At the beginning of the test, replicates A-F were inoculated with 1.0 mL of algal concentrate containing approximately 1.0×10^6 cells/mL. This provided approximately 1.0×10^4 cells/mL at initiation. Replicate G also received an aliquot of the algal concentrate to achieve an initial cell density of 1.0×10^4 cells/mL. Flasks were placed on an orbital shaker table (100 rpm) in a temperature controlled environmental chamber (24±2°C) under continuous cool-white fluorescent lighting. Positions were established by random assignment, and were re-randomized on a daily basis throughout the 4-day test.

At 24, 48, 72, and 96 hours, cell density was measured in each treatment group by direct microscopic counting using a hemacytometer. For the control group, samples from replicates A – F were counted. For all naphthenic acid WAF treatments, samples from replicates A – C were counted. Temperature and pH were measured in all parent solutions prior to distribution of the solutions to the test flasks. At 72 hours, temperature and pH were measured in all replicate G vessels. At 96 hours, temperature and pH were measured in all replicate A vessels.

Measurements of the concentrations of dissolved naphthenic acids in the WAFs were made on samples taken at 0, 72, and 96 hours. The method of analysis included aqueous sample extraction by methylene chloride with detection by Fourier transformed Infrared spectroscopy. The minimum quantifiable limit (MQL) for the method was 0.6 mg naphthenic acids/L. Additional characterization of the exposure solutions included analysis by gas chromatography-mass spectroscopy. This method allowed the proportion of dissolved naphthenic acids to be resolved into families of naphthenic acids having similar carbon numbers and ring numbers.

Limit Test:

No

Test Results

NOEC/LOEC/NOELR/LOELR

	Exposure Duration:	Exposure Units:	Value Description:	Value or Lower Range:	Upper Range:	Units:	Basis for Concentration:
NOELR:	72	Hours	=	10		mg/L	nominal
LOELR	72	Hours	=	20		mg/L	nominal
NOEC	72	Hours	=	7.38		mg/L	arithmetic mean measured
LOEC	72	Hours	=	15.0		mg/L	arithmetic mean measured
NOELR	96	Hours	=	10		mg/L	nominal
LOELR:	96	Hours	=	20		mg/L	nominal
NOEC:	96	Hours	=	7.41		mg/L	arithmetic mean measured
LOEC:	96	Hours	=	14.8		mg/L	arithmetic mean measured

LC/EC/IC/EL/LL Mean Value

Id Naphthenic Acids

Date May 15, 2012

Exposure Duration:	Exposure Units:	Туре	% :	Value Description:	Mean Value or Lower Mean Value:	Upper Mean Value:	Units:	Basis for Effect:	Basis for Concentration:
72	Hours	EL	50	=	41.3		mg/L	Growth Rate	nominal
72	Hours	EL	50	=	23.8		mg/L	Cell Yield	nominal
72	Hours	EC	50	=	29.6		mg/L	Growth Rate	arithmetic mean measured
72	Hours	EC	50	=	17.7		mg/L	Cell Yield	arithmetic mean measured
96	Hours	EL	50	=	43.3		mg/L	Growth Rate	nominal
96	Hours	EL	50	=	24.8		mg/L	Cell Yield	nominal
96	Hours	EC	50	=	29.9		mg/L	Growth Rate	arithmetic mean measured
96	Hours	EC	50	=	18.1		mg/L	Cell Yield	arithmetic mean measured

The NOELR/LOELR and NOEC/LOEC at 72 and 96 hours were the same values when based on growth rate or cell yield.

Algal cells appeared normal with no unusual cell shapes, color differences, flocculation, adherence of algae to the test chambers, or aggregations of algal cells.

Results Remarks:

Concentrations of dissolved naphthenic acids in the test solutions remained stable over the renewal period. The measured concentrations in the solutions at 72 hours were at least 80% of the initial measured concentrations. At 96 hours, the measured concentrations were at least 85% of the initial measured concentrations.

Analysis by GC-MS for carbon number and ring distribution indicated 81-94% of the dissolved naphthenic acids contained 10 to 16 carbon atoms with a prevalence of one and two ring naphthenic acid isomers.

Reliability/Data Quality

Reliability: 1

Reliability Remarks: Reliable without restrictions

Key Study Sponsor

Indicator: Key

Reference

Reference:

Reference:

Report no. 64405, Analytical Bio-Chemistry Laboratories, Columbia, Missouri.

Acute Toxicity to Aquatic Plants

Category Name RECLAIMED SUBSTANCES: Naphthenic acids

Category Chemical: 1338-24-5

Test Substance: 1338-24-5

Test Substance Purity/Composition and Other Test

Id Naphthenic AcidsDate May 15, 2012

Substance Comments :			
Category Chemical Result Type :	unknown		
Test Substance Result Type:	measured		
Method			
Year Study Performed :	1966		
Method/Guideline Followed:	unknown		
Deviations from Method/Guideline :			
Species:	Navicula seminulum		
GLP:	no data		
Analytical Monitoring :	no data		
Test Type:	no data		
Test Vessel:	no data		
Water Media Type:	freshwater		
Test Concentrations:	nominal		
Nominal and Measured Concentrations:			
Total Exposure Period:	96 hours		
Vehicle Us	ed:		
Vehicle Na	me:		
Vehicle An	nount and Units:		
Alkalinity:			
Dissolved	Oxygen:		
pH Value:		Value or Lower Range : Upper Range :	
Test Temp	erature	Value or	

Id Naphthenic AcidsDate May 15, 2012

					Ul	pper	r Range	:					
		Photo (Light/Dark):											
	Salinity:												
	TOC:												
	Water Har	dness	i:		Lov		or Range: Range:						
Test Cond Remarks	:												_
Limit Tes													
Test Re	esuits				NOEC /I OE	C / N	IOEL D	/I OEI					
					NOEC/LOE	C/N			-K				
	Exposure Duration:		posu Inits		Value Description	on:	Value Low Rang	er		oper nge:	Units:	Со	Basis for ncentration:
NOEC:													
LOEC:													
NOELR:													
LOELR:													
				L	C/EC/IC/	EL/	LL Mea	n Val	ue				
Exposu Duratio	•	Туре	% :		Value scription:	Va Lo	lean lue or ower lean alue:	Up _l Me Val	an	Units:	Basis for Effect		Basis for Concentration:
L		naph	then	ic aci	dpoint value ds, CAS 13 red from th	38-2	24-5. Th	ne orig	ginal s	source (of data c	ould r	not be obtained
A total of 12 endpoints were reported in the ECOTOX database. All were be hour tests evaluated on the basis of population growth rate.					ere based on 96-								
		fresh	wate	er dia		ıla s	seminulu						opulations of the EC50 for growth
Reliabi	lity/Data Q	uality	<i>,</i>										
Reliabilit	y:	4 (no	t ass	signa	ble)								
	y Remarks: y Sponsor				reported in d to evalua								nal report could
Indicator Refere	1												

Id Naphthenic Acids

Date May 15, 2012

Reference:

The sensitivity of aquatic life to certain chemicals commonly found in industrial wastes. Final Report No. RG-3965(C2R1), US Public Health Service Grant, Acad. of

Nat. Sci., Philadelphia, PA. 89 p.

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

Memo : Effect of naphthenic acids on survival of bluegill (Lepomis macrochirus)

Remark : The value was reported in a summarized journal article

(Cairns et al., 1965) as originating in Cairns and Scheier

(1962).

Result : 48-hour TLm = 5.6 mg/l naphthenic acids

Reliability : (3) invalid

The endpoint was cited in the text of a journal article

without details of the test.

Reference (6) (7)

Memo : Effect of naphthenic acids on survival of bluegill (Lepomis macrochirus)

Remark: Test chambers were 30x60x30 cm all-glass vessels. Dilution

water was well water. Testing was performed at a temperature of 22 ± 1°C under a 16-h light/8-h dark

photoperiod.

The test included five concentrations of the test substance and a dilution water control. Each test level included 20 fish distributed 10 each to two replicate chambers per

Id Naphthenic Acids

Date May 15, 2012

treatment.

Dissolved oxygen ranged from 4.3 to 8.1 mg/l, pH ranged from 7.4 to 8.0, and temperature ranged from 22 to 24 °C when measured daily during the test. Specific conductance between the test solutions remained constant at 550 (no units given) when measured at the beginning of the test.

The report stated that serial dilutions of the test product were created for testing, although no details were given as to how the serial dilutions or the original solution was created. The raw data indicated that concentrations were expressed as a percent, while the LC50 and confidence interval was reported as parts per million. There was no explanation how the values for percent were related to parts per million.

Critical details of testing procedures and animal culture were omitted from the report.

Result Reference : 96-hour LC50 = 0.0026 mg/l

(14)

Memo

: Effect of naphthenic acids on survival of zebra fish (Brachydanio rerio) embryos

Remark

: Zebra fish embryos were exposed for 48 hours to a range of naphthenic acids concentrations to determine the TLm (median tolerance limit) for embryo survival. Embryos were collected from a culture unit once they attained Stage 21 as designated by Hisaoka and Battle (1958). Ten embryos were exposed to each test solution and control in petri dishes holding 45 ml of the exposure solutions. Exposure solutions were prepared by diluting a stock solution of naphthenic acids (100 mg naphthenic acids in 50 ml acetone) with water. In addition to a control group, nine concentrations of naphthenic acids were prepared at 2.4, 3.2, 4.2, 6.5, 10, 15.5, 24, 32, and 42 ppm naphthenic acids. Mortality was assessed at 24 and 48 hours of exposure. The embryo was considered dead if it had an opaque appearance.

A TLm of 3.5 ppm was obtained by plotting the survival versus concentration on semilog paper and interpolating the 50% survival concentration. The following dose response was given:

Test	Percen
Concentration, ppm	Dead
0 (control)	0
2.4	0
3.2	40
4.2	70
6.5	100
10	100
15.5	100
24	100
32	100
42	100

Reliability

: (2) valid with restrictions

Although the test was conducted prior to the time of standardized test methods, the report provided sufficient information on the dose-response pattern for the test substance.

4. Ecotoxicity	Naphthenic Acids May 15, 2012
Reference	(6) (20)
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5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : = 5880 mg/kg bw

Species : Rat
Strain : Wistar
Sex : Male
Number of animals : 5

Vehicle : other: None, administered undiluted **Doses** : 1, 1.47, 2.15, 3.16, 4.64, 6.81 & 10 g/kg

Method

Year : 1979 GLP : no data

Test substance : other TS: MRD-79-10 (Raw naphthenic acid derived from kerosene) [CAS

number 1338-24-8]

Method: Seven groups of 5 male rats were dosed at 1.0, 1.47, 2.15,

 $3.16,\,4.64,\,6.81,\,$ and $10\,$ g/kg of body weights. Food and water were freely available except for the 16-20 hours prior

to dosing.

The rats were observed 1,2,4, and 6 hours after dosing and

once daily for 14 days. Mortality, toxicity and

pharmacological effects were recorded. Body weights were recorded pretest and in the survivors at 14 days. At 14 days the survivors were sacrificed. All animals were

examined for gross pathology.

Result: Deaths occurred at the four highest dose levels: 3.26, 4.64,

6.81, and 10 g/kg bw. 8/10 animals died at the two highest dose levels. Significant predeath toxic signs included tremors, lethargy, ptosis, ataxia, prostration, negative righting reflex, flaccid muscle tone, piloerection,

diarrhea, chromodacryorrhea, dyspnea and chromorhinorrhea.

Body weight changes were noted in the survivors. Significant necropsy findings in the animals that died

during the study included dilated hearts and

gastrointestinal irregularities.

The LD50 was determined to be 5.88 (4.31-8.02) g/kg bw

Reliability : (2) valid with restrictions

Appears to be comparable to a guideline study with adequate experimental details provided; although the investigators used male rats only, there is sufficient experimental detail to make a conclusion on the study's validity, and the results can be used to assess the potential acute toxicity

of naphthenic acid.

Reference (12)

Type : LD50

Value

Species : Rat

Strain : other: No information

Sex : no data

Number of animals

Vehicle : other: None - administered undiluted

Doses : Method :

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Year : 1955 GLP : no data

Test substance :

Method : "The LD50 ..was determined in rats by use of screening test

procedures similar to those of Smyth and Carpenter." (Smyth, H.F., and C.P. Carpenter. 1944. Place of the range finding test in the industrial toxicology laboratory. J. Indust.

Hyg. & Tox. 26: 269.

Number of animals: "Sufficient animals ...so the the LD50 dose could be computed by either the Weil or the Litchfield and

Wilcoxon method"

Result: Death appears to result from gastrointestinal disturbances,

with the mortality peak occurring on the third to fourth day after administration. The animals exhibited anorexia,

inanition, diarrhea, and asthenia.

The LD50s were determined to be 3.0 g/kg bw (fraction from crude kerosene acids) and 5.2 g/kg bw (fraction from mixed

crude oils)

Test substance: No CAS number identified

1) 7-93% Naphthenic acid fraction from crude kerosene acids 2) 65-69% Naphthenic acid fraction from mixed crude oils

Reliability : (2) valid with restrictions

Although not a guideline or GLP study, and some of the experimental details are not available, the study does appear to be well-conducted, and cites that the investigators followed published methodologies for conducting a statistically valid LD50. The data are

supportive of other acute toxicity studies reported by Exxon

and Pennisi.

Reference (28)

Type : LD50

Value : = 3550 mg/kg bw

Species : Mouse

Strain : other: White - no other information

Sex : Male

Number of animals : Vehicle : Doses : Method : Year

Year : 1977 **GLP** : No

Test substance : other TS: Naphthenic acid - no further information [Assocoiated with CAS

number 1338-24-5 in Toxline search]

Result : Oral administration resulted in 1) CNS depression without

analgesia and no loss of corneal reflex, 2) corneal eye opacity, 3) dryness of mouth, 4) convulsions, 5) diarrhea,

and 6) death due to respiratory arrest.

Reliability : (4) not assignable

This information is taken from a published, meeting abstract. The level of experimental details provided is not

sufficient to verify the conclusions.

Reference (27)

Type: other: Acute oral toxicity study (Not LD50)

Value

Species : Rat
Strain : Wistar
Sex : male/female

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Number of animals : 10 Vehicle : Water

Doses : Method :

Year : 2002 GLP : no data

Test substance :

Method : Female rats were given a single oral dose of naphthenic

acids at 3, 30 or 300 mg/kg bw, while male rats received 300 mg/kg. Control animals were given tap water. All animals were monitored continuously for 12 hr after dosing, and thereafter daily. Changes in body weight, food and water consumption and behavioral or clinical signs were recorded. Following euthanization the liver, kidney, spleen, heart, lung and ovaries were removed, weighed and fixed for

microscopic examination.

Statistical analysis was performed by using a one-way ANOVA to compare means of female dose and control groups with respect to consumption, body weights, and organ weights. A pair wise multiple comparison test was then used in cases where statistical significance was reached. For the male dose and control groups, a Student's t-test was used to compare group means. Probability values of p < 0.05 was

considered statistically significant.

Result: The following effects were seen in the high dose groups:

Decreased food consumption immediately following dosing.

Lethargy and mild ataxia (2/10 females, 3/10 males)

Statistically significant increase relative organ weights:

ovaries, spleen in females- testes, heart in males

7/10 females and 6/10 males exhibiting eosinophilic

pericholangitis

6/10 males and 2/10 females with brain hemorrhage.

The following effects were seen in the mid dose group:

7/10 females and 4/10 males with heart lesions.

Test substance : Naphthenic acid in aqueous solutions (analyzed by mass

spectrometry) containing 55,080, 5508 or 550.0 mg/l naphthenic acids - derived from athabasca sands sands

tailings. [Associated with CAS number 1338-24-5 in Toxline search]

Reliability : (2) valid with restrictions

The study is not an acute toxicity study as defined by OECD SIDS/HPV, however it appears to be well conducted and provides additional information regarding potential acute, non-lethal effects of naphthenic acids following oral

exposure.

Reference (30)

5.1.2 ACUTE INHALATION TOXICITY

5.1.3 ACUTE DERMAL TOXICITY

Type : other: LD50 with irritation Value : > 31600 mg/kg bw

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Species : Rabbit

Strain : New Zealand white Sex : male/female

Number of animals : 2

Vehicle : other: None - administered undiluted

Doses : 3.16 mg/kg

Method :

Year : 1979 GLP : no data

Test substance : other TS: MRD-79-10 (Raw naphthenic acid derived from kerosene) [CAS

number 1338-24-8]

Method : 3.16 g/kg naphthenic acid was applied dermally to the

clipped abraded abdomens of each animal. The area was covered with gauze and secured by a thick plastic binder, which was removed after 24 hours, and the skin washed with

water or corn oil.

According to experimental protocol, no deaths occurred at the initial level, no addition animals were dosed. If one animal died, the experiment was to be repeated using 3 more

groups of animals dosed at varying levels.

Following the skin wash, animals were observed for mortality

and toxic effects at 2 hr and 4 hr, and once daily thereafter. Body weights were recorded pretest and at termination. Dermal irritation was recorded at 24 hr, 3, 7,

10 and 14 days.

The rats were observed 1,2,4, and 6 hours after dosing and

once daily for 14 days. Mortality, toxicity and

pharmacological effects were recorded. Body weights were recorded pretest and in the survivors at 14 days. At 14 days the survivors were sacrificed. All animals were

examined for gross pathology.

Result : No deaths occurred at the 3.16 mg/kg dose level. Most of

the animals (3/4) appeared normal during the first 2 to 4 hours of dosing, after which symptoms of toxicity were observed. 3 out of 4 animals (1 male, 2 female) showed signs of toxicity until day 12 or 13. During the first 5 days, all animals displayed one or more of the following symptoms: lethargy, diarrhea, ptosis, adipsia, anorexia, and

few feces.

The LD50 was determined to be greater than 3.16 g/kg bw Redness and irritation scores were recorded at 24 hr, 3, 7, 10 and 14 days post-washing.

4 Hour occluded sites (DOT, OECD methods)

Mean values (24, 48 & 72 hours) for erythema and edema at

the intact sites were 1.69 and 1.3 respectively.

The initial response of the skin to the test material was slight, with little difference in response between intact or

abraded sites.

The material was judged to be moderately to severely irritating to the occluded skin.

Actual scores were:

Erythema/Eschar Scores

Animal

Numl	oer	1 day	3 day	7 day	10 day 14 day
1M	2	2	4	4	1
2M	1	2	4	4	1
3F	2	4	4	4	0
4F	2	3	4	4	0

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Note: All animals showed signs of scar formation after 14

days.

Edema Animal

Number 1 day 3 day 7 day 10 day 14 day

1M 2 2 2 2M 3 2 2 0 3F 3 2 2 0 3 4F 3 3 2 2 0

Reliability : (1) valid without restriction

Although no indication that it is a GLP study, sufficient

detail is provided to make a conclusion about its validity.

Reference (11)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

5.2.2 EYE IRRITATION

Species : Rabbit
Concentration : Undiluted
Dose : .1 ml

Exposure time

Comment

Number of animals : 3
Vehicle : None
Result : Irritating

Classification Method

Year : 1979 GLP : no data

Test substance : other TS: MRD-79-10 (Raw naphthenic acid derived from kerosene) [CAS

number 1338-24-8]

Method : 0.1 ml naphthenic acid was placed into the conjunctival sac

of eye of each of the six rabbits. The lids were held together briefly to insure adequate distribution. The

untreated eye served as a control.

The rabbits were observed at 1 and 4 hours, and on days 1,

2, 3, 4, and day 7. If a positive score (any score for iritis or opacity, or a score of 2 or more for redness or chemosis) was noted on day 7, ocular reactions were scored on day 10. Likewise readings on day 14 were given if there was a positive reaction on day 10. Fluorescein was used in examining ocular reactions on day 3 and after. The Draize

technique was used as the scoring system.

Result: The following is a summary of data taken from the report:

One animal had a positive corneal score that was noted on days 1 and 2. One animal had a positive iris score which was noted during hours 1 and 4. All animals exhibited positive conjunctival scores at some pint during the first three days of observation. By day 4, no animals showed

positive scores.

The material was judged to be an irritant. (According to

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Draize chart, 4 to 6 rabbits with positive scores observed at 24, 48 or 72 hours). In a later Exxon summary report, eye irritation was judged to be moderate (Exxon, 1980).

Reliability : (1) valid without restriction

Although no indication that it is a GLP study, sufficient

detail is provided to make a conclusion about its validity.

Reference (13)

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type : Sub-chronic

Species: RatSex: FemaleStrain: WistarRoute of admin.: GavageExposure period: 90 Days

Frequency of treatm. : daily, 5 days/week for 90 days

Post exposure period :

Doses : 0.6, 6 & 60 mg/kg **Control group** : yes, concurrent vehicle

Method :

Year : 2002 GLP : no data

Test substance :

Method: Female rats were administered naphthenic acid (orally) at

doses of 0.6, 6, or 60 mg/kg/day, 5 days per week for 90 days. Control animals were given 7 ml tap water. All animals were monitored daily. Changes in body weight, food and water consumption and behavioral or clinical signs were recorded. Blood samples were collected from the ventral tail vein on day 45 of dosing and analyzed for plasma biochemical and hematological effects. Similarly, blood samples taken via cardiac puncture on day 91 were analyzed. Following euthanization the liver, kidney, spleen, heart, lung and ovaries were removed, weighed and fixed for microscopic examination.

Statistical analysis was performed by using a one-way ANOVA to compare group means for consumption, plasma biochemical/

hematological parameters, and organ weights, while a one-way repeated measure ANOVA was used to compare body

weight trends. Probability values of p < 0.05 was

considered statistically significant.

Result: The following significant effects were seen in the high dose groups:

Decreased food consumption immediately following

dosing.

Severe, clonic seizures lasting 20 sec (25%) of animals, observed after day 40 - after which all animals, except one that

died, resumed normal activity.*

Lower mean body weight throughout the exposure period.

Increased relative organ weights: liver, kidney and brain

Reduction in plasma cholesterol on days 45 and 91 (41

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and 43%), Increase in amylase activity on day

45 and 91 (33 and

30%)

Less pronounced differences in total protein concentration (increased) and albumin/globulin ratio (decreased)

5/12 rats with increased glycogen storage.

The following effects were seen in the mid-dose group:

Severe, clonic seizures lasting 20 sec (17%) of animals, observed after day 40 - after which all animals except one that died, resumed normal activity.*

3/12 rats with increased glycogen accumulation

The following effects were seen in the low-dose group:

2/12 rats with increased glycogen accumulation

*Note: Rats in the low-dose (8%) and control (17%) demonstrated milder episodes, characterized primarily by muscle twitching.

Dose-related changes in liver tissue with respect to

glycogen accumulation.

Test substance: Naphthenic acid in aqueous solutions (analyzed by mass

spectrometry) containing 8549, 845.9 or 84.50 mg/l

naphthenic acids derived from Athabasca sands tailings. [Associated with

CAS number 1338-24-5 in Toxline search]

Reliability : (2) valid with restrictions

The study is not a typical subchronic toxicity study as

defined by OECD SIDS/HPV, i.e., the study was conducted with female rats only and examined a limited number of organs. However, it is well-conducted and provides limited information regarding potential subchronic effects of

naphthenic acids following oral exposure.

Reference (30)

Type : Sub-chronic

Species: RatSex: MaleStrain: WistarRoute of admin.: GavageExposure period: 30 daysFrequency of treatm.: Daily

Post exposure period

Doses: 1000 mg/kg bw (no information on number of animals per dose)

Control group : no data specified

Method

Year : 1977 **GLP** : No

Test substance: other TS: Naphthenic acid - no further information [Associated with CAS

number 1338-24-5 in Toxline search]

Method : Male rats were given daily oral doses of 1000 mg/kg

naphthenic acids. No other experimental details provided in

abstract.

Result : The following statements appeared in the abstract:

Repeated daily administration (30 days) of naphthenic acid

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at doses of 1000 mg/kg orally .. revealed a few cases of (1) CNS depression without analgesia and no loss of the corneal reflex (2) hematological changes, (3) weight loss leading eventually to death due to respiratory arrest, (4) gross morphological changes in the liver and stomach, and (5) histomorphological changes in a few selected organs.

Reliability

: (4) not assignable

This information is taken from an abstract. The protocol of the study does not appear to be comparable to a guideline study, and the level of detail is insufficient to judge its

validity.

Reference (27)

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Repeated-Dose Toxicity							
TEST SUBSTANCE	TEST SUBSTANCE						
Category Chemical:	1338-24-5 Naphthenic acids						
Test Substance:	1338-24-5 Naphthenic acids						
Test Substance Purity/Composition and Other Test Substance Comments:	The test sample used in the current program was a blend of naphthenic acids from three sources. The samples were dried under a stream of nitrogen and then re-dissolved in 0.5 mL dichloromethane. The samples were analyzed by GC-MS (Young et al., 2008) and the total ion current mass spectra were collected and tablulated (Holowenko et al., 2002). Based on these data it was determined that there were no significant differences among these samples (Fedorak, 2009). The data indicated that the test material contained constituents with carbon numbers predominantly in the range of C6-C16 (corresponding to a molecular weight range of approximately 116-250) and with a ring distribution of approximately 0 rings (24%), 1 ring (39%), 2 rings (31%), 3 rings (5%) and 4 rings (1%).						
Category Chemical Result Type:	Measured						
Unable to Measure or Estimate Justification:	N/A						
METHOD							
Route of Administration:	Oral						
Other Route of Administration:	N/A						
Type of Exposure:	Gavage						
Species:	Rat						
Other Species:	N/A						
Mammalian Strain:	Sprague-Dawley						

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Other Strain:	N/A
Gender:	Male/female
Number of Animals per Dose:	12/sex/dose group
Concentration:	The naphthenic acids were suspended in corn oil to the appropriate concentrations and administered in 10 ml/kg doses.
Dose:	100, 300, 900 mg/kg/day
Year Study Performed:	2010
Method/Guideline Followed:	OPPTS 870.3650, 2000/OECD 422
GLP:	Yes. Code of Federal Regulations, Title 21, Volume 1, Part 58. Good Laboratory Practice for Nonclinical Laboratory Studies, revised April 1, 2007. OECD. Guideline for the Testing of Chemicals, Section: Health Effects, Subsection 474. Updated and adopted 21 July, 1997.
Exposure Period:	Value or Lower Exposure Duration: Male dosing was for 28-29 days Upper Exposure Duration: Depending on the time at which mating occurred, females were dosed for 39-53 days
Frequency of Treatment:	Daily
Post-Exposure Period:	None

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All rats were examined twice daily for mortality and general health. Detailed physical examinations of all animals were conducted weekly. Additionally, all animals were examined approximately 1 hour after each treatment, and all unusual observations were recorded.

Body weights of male rats were recorded one week prior to test substance administration, on the first day of dose administration, on a weekly basis during the study and at termination. Body weights of female rats were recorded once week prior to test substance administration, on the first day of dose administration and weekly until evidence of copulation was obtained. From that point body weights of female rats were recorded on gestation days (GD) 0, 4, 7, 11, 14, 17, and 20 and on lactation days (LD) 0, 1 and 4 (termination). For females for which there was no evidence of copulation, body weights were recorded weekly until termination. Body weights of offspring were recorded on post-natal day (PND) 1 and then on PND 4, prior to termination. Food consumption by adult animals was also recorded on the same schedule as the body weights.

The potential for nervous system effects was assessed using a functional observation battery (FOB). All rats in the vehicle (corn oil) and naphthenic acid-treated groups were examined prior to dosing, after approximately 28 days of dosing, and, for females, prior to termination. The FOB procedures were based on previously developed protocols (Gad, 1982; Haggerty, 1989; Irwin, 1968; Moser et al., 1988; 1991; O'Donoghue, 1989). The testing was conducted in a sound attenuated room with a white noise generator set to operate at 70 ± 10 dB. The investigators conducting the FOB were not aware of the treatment groups from which the respective animals were taken. The FOB consisted of the following: home cage observations; handling observations; open field observations; sensory observations and neuromuscular observations (Table 2). In addition there were physiological observations including body weight, body temperature and examination for catalepsy. There was also an assessment of locomotor activity which was measured electronically using a computer-controlled system with a series of infrared photobeams in a clear plastic rectangular cage. Animals were tested separately in 60 minute sessions divided into 5 minute intervals.

Method/Guideline and Test Condition Remarks:

On the day of scheduled termination, blood samples were taken from all rats in the corn oil (vehicle) and naphthenic acid-treated rats for assessment of hematological and serum chemistry parameters. The hematological investigation included measurements of total leukocyte count, erythrocyte count, hematocrit, mean corpuscular volume, mean corpuscular hemoglobin, mean corpuscular hemoglobin concentration, platelet count, prothrombin time, activated partial prothrombin time, reticulocyte count, mean platelet volume, red cell distribution width, hemoglobin distribution width, differential leukocyte count, and red cell morphology. The serum chemistry investigation included measurements of concentrations of albumin, total protein, globulin, albumin/globulin ratio, total bilirubin, urea nitrogen, creatinine, alkaline phosphatase, alanine aminotransferase, aspartate aminostransferase, gamma glutamyltransferase, glucose, total cholesterol, calcium, chloride, phosphorus, potassium, sodium, triglycerides, and bile acids.

At termination rats were euthanized by carbon dioxide inhalation. Necropsies were conducted on all animals sacrificed *in extremis* or at study termination. Organs were removed weighed if this was planned, and placed in 10% neutral buffered formalin for histologic examination. The disposition of organs and tissues was as listed in the table below. Note that the target organ investigation encompassed male and female reproductive organs to assist in the assessment of potential reproductive effects.

Tissues collected for weights and/or histological evaluation

Tissue Collected Weight Pathological Examination Adrenals Yes Yes

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Aorta	No	Yes
Bone with marrow (sternebrae)	No	Yes
Bone Marrow Smear	No	Yes
Brain (Cerebrum, Cerebellum)	Yes	Yes
Coagulating Gland	No	Yes
Eyes with Optic Nerve	No	Yes
Esophagus	No	Yes
Stomach	No	Yes
Duodenum	No	Yes
Jejunum	No	Yes
Ileum	No	Yes
Cecum	No	Yes
Colon	No	Yes
Rectum	No	Yes
Heart	Yes	Yes
Kidneys (2)	Yes	Yes
Left femur	No	Yes
Liver (2 lobes)	Yes	Yes
Lungs (fixed by inflation)	Yes	Yes
Lymph nodes (axillary, mesenteric, mar	ndibular)	No Yes
Ovaries with oviducts	Yes	Yes
Pancreas	No	Yes
Peripheral nerve (sciatic)	No	Yes
Pituitary	No	Yes
Prostate	No	Yes
Salivary glands	No	Yes
Seminal Vesicles	No	Yes
Skeletal Muscle (rectus femoris)	No	Yes
Skin with mammary gland	No	Yes
Spinal cord (cervical)	No	Yes
Spleen	Yes	Yes
Testes with epididymides	Yes	Yes
Thymus	Yes	Yes
Thyroids (with parathyroids)	Yes	Yes
Trachea	No	Yes
Urinary Bladder	No	Yes
Uterus with Cervix and Vagina	Yes	Yes
Gross Lesions	No	Yes

Mean parental body weights (weekly, gestation and lactation), body weight changes and food consumption, body weight changes, absolute and relative organ weights, clinical pathology values (except for gamma glutamyltransferase), and continuous FOB values were evaluated by one-way analysis of variance (ANOVA) (Snedecor and Cochran, 1980) to determine intergroup differences between the vehicle control and test substance-treated groups. If the ANOVA revealed significant (p < 0.05) intergroup variance, Dunnett test (Dunnett, 1964) was used to compare the test substance-treated groups to the control group. Histopathological findings in the test substance-treated groups and FOB parameters yielding scalar or descriptive data were compared to the vehicle control group



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using Fisher's Exact Test (Steel and Torrie, 1980). Gamma glutamyltransferase data were evaluated using the Kruskal-Wallis nonparametric ANOVA (Kruskal and Wallis, 1952) to determine intergroup differences between the vehicle control and test substance-treated groups. If the ANOVA revealed significant (p < 0.05) intergroup variance, Dunn Test (Dunn, 1964) was used to compare the test substance-treated groups to the vehicle control group.

TEST RESULTS

Concentration (LOAEL/LOAEC/NOAEL/NOAEC)

Туре	Population: Va	lue Description	: Value or Lower C	oncentration:	Upper Concentration:
NOAEL	Male Sprague-Dawley I	Rats	Systemic Toxicity	100	Mg/kg/day
NOAEL	Female Sprague-Dawle	y Rats	Systemic Toxicity	100	Mg/kg/day
NOAEL	Male Sprague-Dawley r	ats	Neurotoxicity	900	Mg/kg/day
NOAEL	Female Sprague-Dawle	y rats	Neurotoxicity	900	Mg/kg/day

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Two high dose females were terminated on LD 2; one was sacrificed *in extremis* due to acute inflammation of the uterus; the other was sacrificed due to total litter loss. All other rats survived to scheduled termination. Clinical observations, which were noted only in high dose group females and approximately an hour of dosing included hunched posture; rocking, lurching, and/or swaying while ambulating; walking on tiptoes; hypoactivity; and shallow respiration. Some of the high dose group males also exhibited hunched posture.

Body weight gain was reduced in high dose group males but the overall difference was less than 10% and the differences were not statistically significant. Among the females, the body weight gain in the high dose group was approximately 4% below control values but not significantly different at the end of the mating period. These differences in weight gain were associated with significantly reduced food consumption in the high dose group animals.

There were no statistically significant differences in parameters assessed as part of the functional observation battery including home cage observations, handling parameters, open field observations, sensory observations or neuromuscular observations. There were some small differences in body weight gain as indicated previously but other physiological parameters (catalepsy, body temperature) were not affected by treatment. There were also no differences in locomotor activity patterns (data not shown).

There were some hematology changes, primarily reductions in parameters related to hemoglobin content which were considered to have been treatment related. However, as is apparent from Table 1, the differences were small and there was no consistency between males and females.

The clinical chemistry values showed a similar pattern. Among males the only statistically significant differences between control were for creatinine (control value = 0.3 ± 0.1 mg/dL versus a value of 0.4 ± 0.0 in the high dose group, p < 0.01), and chloride (control value = 104 ± 1.1 mEq/L versus a value of 102 ± 1.3 in the high dose group, p < 0.01). Among the female rats, statistically significant differences were found for albumin (control = 4.3 ± 0.2 g/dL versus 4.7 ± 0.3 in the high dose group, p < 0.05), total protein (control = 6.3 ± 0.3 g/dL versus 6.7 ± 0.4 in the high dose group, p < 0.05), glucose (control = 115 ± 11 mg/dL versus 130 ± 8.0 in the high dose group, p < 0.05), cholesterol (control = 6.4 ± 0.4 mg/dL in the control versus 8.9 ± 1.9 in the high dose group, p < 0.05), calcium (control = 10.6 ± 0.4 mg/dL in the control versus 11.5 ± 0.6 in the high dose group, p < 0.01), and phosphorus (control = 3.9 ± 0.6 mg/dL versus 5.5 ± 1.2 in the high dose group, p < 0.05). All of the differences were small and within the historical range of the laboratory. Additionally, most were significant at only the 0.05 level, and there was no consistency of response between the sexes. In the absence of any corresponding pathological findings, these differences were most likely incidental.

The only notable gross observations were those of pale kidneys in the high dose males and a reduction in the number of *corpora lutea* in the high dose group females. Otherwise, the results of the gross examination were not remarkable. Organ weight determinations in males revealed significant increases in weights of liver, kidney, thyroid/parathyroid and epididymis although the differences in thyroid/parathyroid and epididymal weights were only statistically different when compared on a relative to body weight basis. In females, there was a significant increase in liver weights and significant reductions in lung weights, and absolute uterine weights (table 3). The lung weights were within the historical range for the laboratory, and were not associated with any pathological changes. The uterine weights were not significantly different when compared relative to body weights. All gravid females were in lactational anaestrus and undergoing involution. Uterine weight values all fell within the historical range for the laboratory and were not associated with any gross, histopathologic or clinical pathology changes. Other than the uterine weights there were no microscopic differences in the reproductive organs of the male and female rats.

Results Remarks:

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The results of the pathological investigation are summarized in Table 3. Kidney changes, reported in male rats only, were consistent with hyaline-droplet nephropathy (a2u-globulin-mediated nephropathy). The liver changes, found in organs from both male and female rats from the high dose group, were described as hepatocellular hypertrophy. Other changes included cortical lymphoid depletion of the thymus in females, primarily in rats from the high dose group. Epithelial hypertrophy and cytoplasmic vacuolation of the thyroid gland was noted in all treated animals, and cytoplasmic vacuolation of the zona fasciculate in the adrenal cortex was reported in males from all treatment groups and in high dose group females. The microscopic examination also revealed minimal cardiomyopathy which occurred with increased incidence in the males in the 100, 300 and 900 mg/kg/day groups. The pathologist noted that cardiomyopathy is a common finding in rats (Greaves, 2007a), that the incidence of cardiomyopathy in the treated animals was within the historical range of the laboratory, and that the severity of cardiomyopathy in the treated male rats was similar to or less than the degree of severity found in the control animals. The pathologist also noted that the cardiomyopathy was not associated with any gross observations, organ weight changes or alterations in clinical pathology parameters.

The gross and pathological assessments did reveal some differences that were treatment-related but were unlikely to have been toxicologically important. Liver weights were significantly increased in high dose groups of both male and female rats, and there was also a statistically significant increase in liver weight in the 300 mg/kg/day dose group in the males. The histological findings were essentially limited to minimal evidence of hepatocellular hypertrophy in the high dose group animals. As none of the liver enzyme markers were increased, this was most likely evidence of enhanced metabolic capacity and adaptive rather than adverse (Cattley and Popp, 2002). Kidney weights were significantly elevated in the male rats from the high dose group, but not in the female rats. The histological evidence revealed the presence of hyaline droplets, mostly judged to have been of minimal severity, which increased in frequency in the male rats in a dose-dependent manner. As these were not found in female rats, the histological findings and gender-specificity, suggest the kidney changes were the consequence of an a-2u-globulin-related process which is male rat specific and not relevant to humans (Hard et al., 2008; Baetcke et al., 1991; Swenberg and McKeeman, 1998).

Minimal cardiomyopathy was reported to have increased in a dose-related fashion in male rats but was not considered to have been toxicologically important. In part because this is a common observation in control rats (Greaves et al., 2007b), and, additionally because the incidence was within the historical control range of the laboratory, the severity was not greater than that seen in the control groups, and because these microscopic observations were not associated with any other gross or clinical findings.

Other changes included higher mean thyroid/parathyroid weights with corresponding epithelial hypertrophy and cytoplasmic vacuolation. The histologic changes were mostly judged as minimal. It is plausible that these changes reflected a compensatory response related to the increased metabolic capacity of the liver and more rapid turnover of thyroid hormones (Curran, 1991; Capen, 1997). Lymphoid depletion of the thymus was observed in the high dose females and microscopic findings of cytoplasmic vacuolation of the adrenal cortex were noted in the males and high dose group females. The lymphoid cortical depletion of the thymus and adrenal cortex vacuolation were considered to have been stress responses (Greaves, 2007b) although cytoplasmic vacuolation of the adrenal cortex can also occur spontaneously (Frith et al., 2000) or as the result of pharmacological effects (Greaves, 2007c). The overall no effect level for all systemic effects was 100 mg/kg/day.

Table 1 Results o

Parameter Measured Corn Oil Control 100 mg/kg/day 300 mg/kg/day 900 mg/kg/day

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Males, data taken at tern	ninal sacrifice				
Red Blood Cell Count (10	⁾⁶ /ul) ^b	9.22 <u>+</u> 0.54	9.28 <u>+</u> 0.28	8.91 <u>+</u> 0.34	8.78
<u>+</u> 0.22 ^c					
Hemoglobin (g/dL) ^b 1	.5.7 <u>+</u> 0.72	15.8 <u>+</u> 0.48	15.2 <u>+</u> 0.54	14.7 <u>+</u> 0.44 ^c	
	8.1 <u>+</u> 2.4	48.6 <u>+</u> 1.6	46.5 <u>+</u> 1.5	45.0 <u>+</u> 1.8 ^d	
	354 <u>+</u> 151	885 <u>+</u> 84	803 <u>+</u> 144	976 <u>+</u> 87 ^d	
Leukocytes, absolute ((1)	0 ³ /ul) ^b	0.02 <u>+</u> 0.02	0.03 <u>+</u> 0.02	0.02 <u>+</u> 0.02	0.04
<u>+</u> 0.03 ^a					
RDW (%) ^b 1	1.4 <u>+</u> 0.4	11.5 <u>+</u> 0.4	11.6 <u>+</u> 0.4	12.5 <u>+</u> 0.6 ^d	
HDW $(g/dL)^b$ 2	.58 <u>+</u> 0.10	2.68 <u>+</u> 0.12	2.76 <u>+</u> 0.16 ^c	2.77 <u>+</u> 0.27 ^c	
Females, data taken at te		day 4)			
White blood cell count ^b 5	5.15 <u>+</u> 1.30	6.89 <u>+</u> 1.58	7.68 <u>+</u> 2.24 ^c	7.59 <u>+</u> 1.85 ^c	
APTT (seconds) ^b 1	.6.8 <u>+</u> 1.9	15.9 <u>+</u> 2.3	15.8 <u>+</u> 3.1	13.9 <u>+</u> 1.4 ^c	
Lymphocytes, absolute (3	10³/ul)	3.32 <u>+</u> 0.61	4.50 <u>+</u> 1.42	5.11 <u>+</u> 1.75 ^c	4.96
<u>+</u> 1.60 ^c					
Monocytes, absolute (10 ³	³ /ul)	0.11 <u>+</u> 0.10	0.24 <u>+</u> 0.21	0.21 <u>+</u> 0.12	0.35
<u>+</u> 0.23 ^c					

- 1. Parameters not affected by treatment included:
 - a. Males white blood cell count, mean corpuscular volume (fL), mean corpuscular hemoglobin (pg), mean corpuscular hemoglobin content (g/dL), prothrombin time (sec), APTT (sec), reticulocytes (%), reticulocytes, absolute (10³/ul), MPV (fL), neutrophils (%), lymphocytes (%), monocytes (%), eosinophils (%), basophils (%), leucocytes(%), neutrophils, absolute (10³/ul), lymphocytes, absolute (10³/ul), monocytes, absolute (10³/ul), eosinophils, absolute (10³/ul), basophils, absolute (10³/ul).
 - b. Females red blood cell count (10⁶/ul), Hemoglobin content (g/dL), hematocrit (%), mean corpuscular volume (fL), mean corpuscular hemoglobin (pg), mean corpusc ular hemoglobin content (g/dL), platelet count (10³/ul), prothrombin time (sec), reticulocytes (%), reticulocytes, absolute (10³/ul), MPV (fL), neutrophils (%), lymphocytes (%), monocytes (%), eosinophils (%), basophils (%), leucocytes(%), neutrophils, absolute (10³/ul), eosinophils, absolute (10³/ul), basophils, absolute (10³/ul), Leukocytes absolute (10³/ul), RDW (%), HDW (g/dL)

Table 2. Statistically significant changes in terminal body weights and organ weights. The data are given as mean \pm SD.

Parameter Males	Sham Control	Corn Oil Control	100 mg/kg/day	300 mg/kg/day	900 mg/kg/day
Final Body Weight	467 <u>+</u> 27	454 <u>+</u> 45	448 <u>+</u> 45	439 ± 34	412 ± 28
Liver	15.61 <u>+</u> 1.43	13.46 <u>+</u> 2.01	13.98 <u>+</u> 2.04	15.69 ± 1.83*	19.94 ± 2.08 ^b
Kidney	3.51 <u>+</u> 0.25	3.21 <u>+</u> 0.20 ^a	3.38 <u>+</u> 0.39	3.53 ± 0.33	3.77 ± 0.46 ^b
Heart	1.46 <u>+</u> 0.09	1.46 <u>+</u> 0.21	1.41 <u>+</u> 0.14	1.43 <u>+</u> 0.13	1.32 <u>+</u> 0.13
Thyroid/parathyroid	0.019 <u>+</u> 0.002	0.019 <u>+</u> 0.001	0.020 <u>+</u> 0.002	0.020 <u>+</u> 0.002	0.020 <u>+</u> 0.002
Epididymis (LT)	0.57 <u>+</u> 0.14	0.60 <u>+</u> 0.05	0.60 <u>+</u> 0.04	0.66 <u>+</u> 0.05 ^a	0.63 <u>+</u> 0.06

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Epididymis (RT)	0.62	2 <u>+</u> 0.04	0.62 <u>+</u> 0.0	6	0.61 <u>+</u> 0.03	0.66 <u>+</u> 0.	04	0.65 <u>+</u> 0.06	5	
Females										
Final body Weight	335	<u>+</u> 25	313 <u>+</u> 23		301 <u>+</u> 30	294 <u>+</u> 24		289 + 24		
Liver		<u></u> 5 <u>+</u> 2.0	11.7 <u>+</u> 1.5		12.1 <u>+</u> 1.1	13.3 <u>+</u> 1.		17.9 <u>+</u> 2.4	b	
Kidney		9 <u>+</u> 0.17	2.07 ± 0.1		2.11 <u>+</u> 0.15	2.05 <u>+</u> 0.		2.17 ± 0.19		
Heart		1 <u>+</u> 0.23	1.10 ± 0.1		1.08 ± 0.10	$1.07 \pm 0.$		1.01 + 0.13		
Lungs		5 <u>+</u> 0.13	1.40 ± 0.1		1.26 ± 0.12°	$1.20 \pm 0.$		1.20 ± 0.07		
Uterus/Vagina		7 <u>+</u> 0.19	1.40 ± 0.1 $1.00 + 0.1$		$0.86 + 0.08^{\circ}$	$0.88 \pm 0.$		0.85 ± 0.12		
a = P < 0.0	05, b = P <	< 0.19	1.00 <u>+</u> 0.1	7	0.00 <u>+</u> 0.00	0.66 <u>+</u> 0.	11	0.05 <u>+</u> 0.12	<u> </u>	
Table 3. Summary	of microsco	opic findinas f	rom rats fol	lowina r	repeated treatme	nt with refir	ned naph	thenic acids		
·				_	·		•			Males
Doses, mg/kg/day		100	300	900	Corn Oil	100	300	900		
N	12	12	12	12	9	12	10	10		
Kidney										
Hyaline Droplets	0	3	10 ^b	11 ^b	0	0	0	0		
Minimal .	0	3	9 ^b	9^{b}	0	0	0	0		
Mild	0	0	1	2						
Nephropathy	0	0	2	9 ^b	0	0	0	0		
Minimal	0	0	2	5 ^a	0	0	0	0		
Mild	0	Ö	0	4	· ·	· ·	Ū	· ·		
	· ·	•	· ·	•						
Liver										
Hypertrophy, hepa	tocellular,	centrilobular	0	0	0	8 ^b	0	0	0	
71 1 77 1 -	10 ^b									
Minimal	0	0	0	8 ^b	0	0	0	10 ^b		
	· ·	•		•	· ·	· ·	Ū			
Vacuolation, hepat	tocellular	2	1	2	0	0	1	0	2	
Minimal	1	1	2	0	Ö	1	0	2		
Mild	1	0	0	0	Ö	Ō	0	0		
	•	J	J	J	v	J	J	J		
Thymus										
Depletion, lympho	id. cortex	0	0	0	0	0	1	0	5 ^a	
Minimal	0	Ö	Ö	Ö	Ö	1	Ō	4	J	
Mild	0	Ö	Ö	0	0	Ō	0	i		
rillu	U	O	O	U	U	U	U	1		
Thyroid										
Hypertrophy, epith	nelial	0	6ª	9^{b}	11 ^b	0	3	4	8^{b}	
Minimal	0	6ª	9 ^b	11 ^b	0	3	4	6ª	-	
Mild	0	0	Õ	0	0	0	0	2		
4	•	•	J	J	Ŭ	•	•	_		
Vacuolation, cytop	lasmic	0	6ª	9 ^b	10 ^b	0	3	4	8 ^b	
Minimal	0	6ª	9 ^b	10 ^b	0	3	4	8 ^b	J	
i iii iii ii ii	5	J	,	10	U	5	-T	U		

5.	Toxicity					•	ohthenic Ac y 15, 2012	ids		
	Adrenal Cortex Vacuolation, cytoplasmic Minimal 0 Mild 0 Heart	2 0 2 0	2 3 0	3 2 0	2 0 0	0 0 0	0 0 0	0 1 1	2	
	Cardiomyopathy		4 8ª	4 8ª	3 0	3 0	2 0	4 0		
Conclusion: RELIABILITY/DATA QU	The overall no effect leve	for all syster	nic effects wa	s 100 mg/kg	g/day.					
Reliability:	1									
Reliability Remarks:	Reliable without restrictio	ns								
Key Study Sponsor Indicator:	Key study for repeated do	se toxicity ar	nd for reprodu	ctive toxicity	/					
REFERENCE										
Reference:	WIL Research (2012). WI Reproduction/Dev Micronucleus Test	elopmental T	oxicity Screer	ning Test of I	Naphthenic A	Acid with a M	oxicity Study ammalian Er	with the ythrocyte		

Date May 15, 2012

5.5 GENETIC TOXICITY 'IN VITRO'

Type : Bacterial reverse mutation assay

System of testing : S. typhimurium TA100, TA1535, TA97, TA98

Test concentration : Not indicated

Cycotoxic concentr. : With and without metabolic activation: >333 µg/plate

Metabolic activation: with and without

Result : Negative

Method : other: US National Toxicology Program protocols

Year : 1993 **GLP** : Yes

Test substance : other TS: Sodium naphthenate [CAS number 61790-13-4] - Study indicates

that it is a C7 naphthenic acid

Remark: Test material is a C7 naphthenic acid, whereas those produced

commercially are mixtures of naphthenic acids predominantly in the C10-C30 range. Consequently, the results of this study are to be used as

supplemental data only.

Reliability : (1) valid without restriction

Reference (24)

Type : Cytogenetic assay

System of testing: Measuring Chromosomal Aberration Frequencies in Chinese Hamster

Ovary Cells (CHO)

Test concentration: Without activation: 54, 116 & 250 μg/ml. With activation: 25, 54, 116 & 250

μg/ml.

Cycotoxic concentr. : Not indicated

Metabolic activation : with and without

Result : Negative

Method : other:US National Toxicology program protocols

Year : 1991 **GLP** : Yes

Test substance : other TS: Sodium naphthenate [CAS number 61790-13-4] - Study indicates

that it is a C7 naphthenic acid

Remark : Solvent control: water

Positive controls:

Without metabolic activation Mitomycin C (0.4 ug/ml)

With metabolic activation Cyclophosphamide (20 ug/ml)

Metabolic activation Arochlor 1254 induced,

Sprague-Dawley male rat

liver S9 fraction

Test material is a C7 naphthenic acid, whereas those produced

commercially are mixtures of naphthenic acids predominantly in the C10-C30 range. Consequently, the results of this study are to be used as

supplemental data only.

Reliability : (1) valid without restriction

Reference (24)

Type : Cytogenetic assay

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Date May 15, 2012

System of testing : Measuring Chromosomal Aberration Frequencies in Chinese Hamster

Ovary Cells (CHO)

Test concentration

Cycotoxic concentr. : Not indicated

Metabolic activation : with and without

Result

Method : other:US National Toxicology program protocols

Year : 1991 **GLP** : Yes

Test substance : other TS: Sodium naphthenate [CAS number 61790-13-4] - Study indicates

that it is a C7 naphthenic acid

Remark : Without activation:

17, 59, 167, 500 ug/ml (Trial 1) 100, 150, 200, 250 ug/ml (Trial 2)

With activation:17, 59, 167, 500 ug/ml

Solvent control: water

Positive controls:

Without metabolic activation - Mitomycin C (0.001 and 0.004

ug/ml)

With metabolic activation - Cyclophosphamide (0.125 and

0.500 ug/ml)

Metabolic activation: Arochlor 1254 induced,

Sprague-Dawley male rat

liver S9 fraction

Test material is a C7 naphthenic acid, whereas those produced

commercially are mixtures of naphthenic acids predominantly in the C10-C30 range. Consequently, the results of this study are to be used as

supplemental data only.

Result : Weakly positive (trial 1- without metabolic activation)

Positive (trial 2 - without metabolic activation)

Negative (with metabolic activation)

Reliability : (1) valid without restriction

Reference (24)

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Date May 15, 2012

5.6 GENETIC TOXICITY 'IN VIVO

Genetic Toxicity <i>in vivo</i>							
TEST SUBSTANCE	TEST SUBSTANCE						
Category Chemical:	1338-24-5 Naphthenic acids						
Test Substance:	1338-24-5 Naphthenic acids						
Test Substance Purity/Composition and Other Test Substance Comments: The test sample used in the current program was a blend of naphthenic acids from three sources. The samples were dried under a stream of nitrogen and then re-dissolved in 0.5 mL dichloromethane. The samples were analyzed by GC-MS (Young et al., 2008) at the total ion current mass spectra were collected and tablulated (Holowenko et al., 2002). Based on these data it was determined that there were no significant differences among these samples (Fedorak, 2009). The data indicated that the test material contained constituents with carbon numbers predominantly in the range of C6-C16 (corresponding to molecular weight range of approximately 116-250) and with a ring distribution of approximately 0 rings (24%), 1 ring (39%), 2 ring (31%), 3 rings (5%) and 4 rings (1%).							
Category Chemical Result Type: Measured							
METHOD							
Type of Study:	In vivo mutagenesis (chromosomal aberrations)						
Type of Test:	Micronucleus Test (OPPTS 870.5395)/OECD 474						
Route of Administration:	Oral gavage						
Species:	Rats						
Strain:	Sprague-Dawley						
Gender:	Male and Female						
Dose:	100, 300, 900 mg/kg/day						

Id Naphthenic Acids

Year Study Performed:	2010
Method/Guideline Followed:	OPPTS 870.5395/OECD 474
GLP:	Yes
Duration of Treatment/Exposure Period and Units:	Approximately 30 days
Frequency of Treatment:	Daily
Positive, Negative and Solvent Control Substance(s):	Sham control (no material administered) Negative control (corn oil) Positive control (cyclophosphamide, 60 mg/kg/day)
Post-Exposure Period:	None
Number of Animals per Sex per Dose:	6 males/6 females
Method/Guideline and Test Condition Remarks:	The micronucleus test was consistent with the US EPA guidelines for studies of this type (OPPTS 870.5395) and with OECD 474. The testing was in accordance with Good Laboratory Practice Guidelines of the OECD (OECD, 1997) and the U.S. EPA (CFR, 2007). Bone marrow was collected from all animals at terminal sacrifice and flushed into a centrifuge tube using a syringe containing heat inactivated fetal bovine serum (HI FBS). The bone marrow was centrifuged, the majority of the HI FBS was decanted, and the pellet was re-suspended. Bone marrow smears were prepared by placing single drops of suspension on microscope slides (minimum of two per preparation). The slides were coded, air dried, fixed in methanol and allowed to air dry a second time. Coded slides were stained with acridine orange (Hayashi et al., 1983). A total of 1000 erythrocytes/slide were evaluated (both polychromatic (PCE) and normochromatic erythrocytes (NCE)), and the PCE/NCE ratio was calculated. The number of micronucleated PCEs from a total of 2000 PCEs was then determined for each animal. The percentages of PCEs , micronucleated cells in NCEs, and the ratios of PCEs to total erythrocytes in the test substance- and vehicle-treated groups were compared using ANOV A (Snedecor and Cochran, 1980). If the ANOVA revealed significant (p < 0.05) intergroup variance, Dunnett Test (Dunnett, 1964) was used to compare each test substance-treated group to the vehicle control group. In addition, the positive control and vehicle control groups were compared using a separate parametric one-way ANOVA (Snedecor and Cochran, 1980).
TEST RESULTS	
Systemic Toxicity:	No treatment-related bone marrow effects.

Id Naphthenic Acids

Genotoxic Effect:	statistically from thos	e in the sham and	vehicle control g	roups. A significant ind	crease in micro	ned naphthenic acids did not differ onucleus frequency was found in materia nat the test had worked as expected.
	Summary of results o Treatment (N=5)	Gender		wing repeated treatmer PCEs/2000 PCEs (N= hrocytes		
	Corn Óil	Males Females	8 8	0.08 <u>+</u> 0.08 0.08 <u>+</u> 0.12	3 4	0.54 <u>+</u> 0.07 0.69 <u>+</u> 0.11
	Sham Control	Males Females	6 8	0.06 ± 0.04 0.08 ± 0.08	3	0.52 ± 0.11 0.55 ± 0.17
	Naphthenic Acid 100 mg/kg/day	Males	7	0.07 <u>+</u> 0.07	1	0.53 <u>+</u> 0.09
Results Remarks:	300 mg/kg/day	Females Males Females	4 4 5	0.04 ± 0.04 0.04 ± 0.04 0.06 + 0.05	7 3 5	0.65 ± 0.16 0.49 ± 0.67 0.67 ± 0.13
	900 mg/kg/day	Males Females	8 5	0.08 <u>+</u> 0.08 0.06 <u>+</u> 0.05	5 5 5	0.61 ± 0.11 0.75 ± 0.19
	Positive Control (Cycl 60 mg/kg/day	ophosphamide) Males	128	1.28 <u>+</u> 0.14 ^a	13	0.40 <u>+</u> 0.21
	a. P < 0.05	Females	97	$0.97 \pm 0.19a$ matic erythrocytes. NC	16	0.51 ± 0.12 ^a
Conclusion:	Naphthenic acids did	not induce chromo	somal aberration	ns under the conditions	of the test	
RELIABILITY/DATA QUALITY						
Reliability:						
Reliability Remarks:	Reliable without restr	ictions				
Key Study Sponsor Indicator:	Key study for in vivo	mutagenic potentia	al			
REFERENCE						

5.	To	xic	ity
			•

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Date May 15, 2012

Reference:

WIL Research (2012). WIL-402011. A Combined 28-Day Repeated Dose Oral (Gavage) Toxicity Study with the Reproduction/Developmental Toxicity Screening Test of Naphthenic Acid with a Mammalian Erythrocyte Micronucleus Test in Rats. WIL Research Laboratories, LLC, Ashland, OH

Date May 15, 2012

5.7 CARCINOGENICITY

Species : Mouse **Sex** : Female

Strain : other: No information available

Route of admin. : Dermal Exposure period : 2 Years Frequency of treatm. : 2 times/day

Post exposure period

Doses : 0.05 ml undiluted

Result

Control group : no data specified

Method :

Year : 1987 GLP : no data

Test substance : other TS: Calcium naphthenate [CAS number 61789-36-4]

Method : Not described; listed in summary as "non-TSCA

Protocol/Guideline (voluntary test)"

Result: The following statements appeared in the abstract:

Clinical observations included mild irritation, hair loss, shiny patches on the skin, and flaking skin surfaces. These progressed to moderate irritation (observed with sores and scabs on the treated site), or severe irritation caused by large sores or visible ulcers. In the negative control group, no cutaneous tumors developed at or distant to treated sites. Twelve epidermal and one dermal tumor at the treated sites were observed in eight mice that were exposed to the test material. Four of the tumors were malignant and none were benign. The first of these neoplasms were reported after 392 days of treatment. No metastatic tumors

were present.

Reliability : (4) not assignable

This information is taken from an EPA site that summarizes results of testing submitted under TSCA. The protocol of

the study does not appear to be comparable a guideline study as indicated in the summary. In addition, the material used (calcium naphthenate) was judged not to be similar to commercially available naphthenic acids.

Consequently, the study is for supplemental use only.

Reference (37)

5.8.1 TOXICITY TO FERTILITY

Type : One generation study

Species : Rabbit
Sex : male/female
Strain : New Zealand white

Route of admin. : Dermal Exposure period : 10 weeks

Frequency of treatm. : 6 hr/day, 5 days/week

Premating exposure period

Male : 10 weeks Female : Not exposed

Duration of test: 10 week exposure period prior to mating, gestation and delivery. Total

duartion of study was approximately 22 weeks

Date May 15, 2012

No. of generation

studies

Doses : Undiluted

: 1

Control group : other: carrier oil as present in the test substance

Method

Year : 1984 GLP : no data

Test substance : other TS: Calcium naphthenate, Shell SAP Oil [CAS number 61789-36-4]

Method : A group of 12 male rabbits was dermally exposed to 2 ml undiluted test

substance or control vehicle for six hours daily for 5 days per week for 10 weeks. Body weights were recorded weekly and at the end of 10 weeks, each male was mated with 2 untreated female rabbits. Half of the males of each group were killed and necropsied after mating. The remaining males were weighed weekly and necropsied approximately 12 weeks later. Macroscopic and microsopic examinations of the male reproductive tracts

were carried out on all rabbits.

The females were necropsied on day 29 of gestation. Numbers of corpora lutea, total implantations, pre-and post-implantation losses and numbers of

viable fetuses were recorded.

Result : All male rabbits survived with the exception of one control that died after 9

weeks of exposure, having shown no unusual clinical signs. There were no

systemic toxicity, application site toxicity,

or statistically significant changes in body weights

observed in the test animals during the 10 week exposure period or the 12 week post-exposure period. In the male animals, there were no significant changes in the testes weights. In the females, there were no significant differences in the number of implantations, or in pre-and post-implantation losses. In addition, there were no differences in viable fetuses to those females that were mated with exposed males compared to those mated with unexposed males. The study also reported that there were no macroscopic or microscopic pathological findings in the male

reproductive tract.

Reliability : (2) valid with restrictions

The study has sufficient detail, however, the protocol does not appear to be comparable to a guideline study. In addition, the material used (calcium naphthenate) was judged not to be similar to commercially available naphthenic acids. Consequently, the study is for supplemental use only.

Reference (32)

Id Naphthenic Acids

Reproductive Toxicity				
TEST SUBSTANCE				
Category Chemical:	1338-24-5 Naphthenic acids			
Test Substance:	1338-24-5 Naphthenic acids			
The test sample used in the current program was a blend of naphthenic acids from three sources. The samples were dried under a stream of nitrogen and then re-dissolved in 0.5 mL dichloromethane. The samples were analyzed by GC-MS (Young al., 2008) and the total ion current mass spectra were collected and tablulated (Holowenko et al., 2002). Test Substance Purity/Composition and Other Test Substance Comments: Based on these data it was determined that there were no significant differences among these samples (Fedorak, 2009). The data indicated that the test material contained constituents with carbon numbers predominantly in the range of C6-C16 (corresponding to a molecular weight range of approximately 116-250) and with a ring distribution of approximately 0 rings (24%), 1 ring (39%), 2 rings (31%), 3 rings (5%) and 4 rings (1%).				
Category Chemical Result Type :	Measured			
Unable to Measure or Estimate Justification :	N/A			
METHOD				
Route of Administration:	Oral			
Other Route of Administration:	N/A			
Type of Exposure:	Gavage			
Species:	Rat			
Other Species:	N/A			
Mammalian Strain:	Sprague-Dawley			
Other Strain:	N/A			

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Gender:	Male/Female
Number of Animals per Dose:	12/sex/dose group
Concentration:	The test materials were suspended in corn oil to the appropriate concentrations and administered daily in bolus doses of 10 ml/kg
Dose:	100, 300, 900 mg/kg/day
Year Study Performed :	2010
Method/Guideline Followed:	OPPTS 870.3650, 2000/OECD 422
GLP:	Yes. The testing was in accordance with Good Laboratory Practice Guidelines of the OECD (OECD, 1997) and the U.S. EPA (CFR, 2007).
Exposure Period:	Value or Lower Exposure Duration: Upper Exposure Duration: Males were exposed 28-29 days Females were exposed for 39-53 days depending on the day on which mating occurred.
Frequency of Treatment:	Daily
Post-Exposure Period:	None
Method/Guideline and Test Condition Remarks:	Dosing of males was initiated 14 days prior to pairing and throughout a 14 day mating period for a total of 28-29 doses. Dosing of females was also initiated 14 days prior to pairing and continued throughout the mating and gestational periods until study termination on post-natal day 3. The total number of doses ranged from 39-53 depending on the time at which mating occurred. Body weights of female rats were recorded once week prior to test substance administration, on the first day of dose administration and weekly until evidence of copulation was obtained. From that point body weights of female rats were recorded on gestation days (GD) 0, 4, 7, 11, 14, 17, and 20 and on lactation days (LD) 0, 1 and 4 (termination). For females for which there was no evidence of copulation, body weights were recorded weekly until termination. Body weights of offspring were recorded on post-natal day (PND) 1 and then on PND 4, prior to termination. Food consumption by adult animals was also recorded on the same schedule as the body weights.

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Mating was initiated after 14 days of dosing. Rats were mated on a 1:1 basis within each treatment group, females were cohoused with the males. Each mating pair was evaluated on a daily basis during the mating period. Successful mating was confirmed by the presence of a vaginal copulatory plug or the presence of sperm in a vaginal lavage. The day on which mating was confirmed was designated as gestational day 0.

All females confirmed to have mated were placed in plastic maternity cages once mating was confirmed. Females for which copulation was not detected were placed in maternity cages at the end of the 14 day mating period. All females were allowed to deliver and to rear their young to post-natal day 4. On the day of parturition, all pups were examined for viability, for the presence of gross malformations and to assess gender. The numbers of live and stillborn pups were recorded. Length of gestation was calculated as the time from confirmation of mating to the onset of delivery. Females for which there was no evidence of mating were sacrificed on post-cohabitation day 25, those that showed evidence of mating but failed to deliver were euthanized on post-mating day 25, and all others were euthanized on post-natal day 4. Uteri with no microscopic evidence of implantation were opened and subsequently placed in 10% ammonium sulfide solution for detection of early implantation loss (Salewski, 1964).

All offspring were uniquely identified and examined daily for signs of mortality and ill health. All offspring were individually weighed on PND 1 and 4. Gender was assessed on PND 0 and 4. At scheduled termination, PND 4, all surviving offspring were euthanized and discarded without further examination.

Parental mating, fertility, conception and copulation indices were analyzed using the Chi-square test with Yates' correction (Hollander and Wolfe, 1999). Mean parental body weights (weekly, gestation and lactation), body weight changes and food consumption, offspring body weights and body weight changes, gestation length, numbers of former implantation sites, numbers of corpora lutea, number of pups born, live litter size on PND 0, unaccounted for sites, absolute and relative organ weights, and pre-coital intervals were evaluated by one-way analysis of variance (ANOVA) (Snedecor and Cochran, 1980) to determine intergroup differences between the vehicle control and test substance-treated groups. If the ANOVA revealed significant (p < 0.05) intergroup variance, Dunnett test (Dunnett, 1964) was used to compare the test substance-treated groups to the control group.

Note that an examination of target organs including male and female reproductive organs was also carried out as part of this test. Organs examined included: ovaries with oviduct, uterus with cervix and vagina, testes with epididymides, prostate and seminal vesicles. The ovaries, testes and uteri were weighed and all were examined histologically. The absolute epididymal weights were increased in the 900 mg/kg/day group but were not significantly different when expressed on a per body weight basis. The uterine weights were also significantly elevated but this was considered to have been a consequence of the fact that the females were all in lactational anaestrous. The uterine weights were within the historical range of the laboratory and were not considered to have been toxicologically important. There were no weight differences in any of the other organs and no pathological changes in any of the reproductive organs at the highest dose tested (900 mg/kg/day).

Pre-Mating Exposure / Males:

14 days

Pre-Mating Exposure / Females:

14 days

TEST RESULTS

Id Naphthenic Acids

Type NOAEL NOAEL NOAEL NOAEL	Population: Male Rats Female Rats Male Rats Female Rats	Value Description: Value of Mating 900 Mating 900 Reproductive Organ Effects Reproductive Organ Effects	900 900 900	n: Upper Co Mg/kg/da Mg/kg/da Mg/kg/d Mg/kg/d	y y lay	iits:
		There was no evidence of treatme differences in frequency of mating Note also as indicated above that Table 1. Summary of reproductive acids.	, time to mate, mating s	success or length of the	e gestational period (table 1).
Results:		Dose (mg/kg/day)Corn Oil Control100 mg/kg/day300 mg/kg/dayNumber of females paired121212Number of female mated121210Number of females pregnant a91210Number of females with litters91210Pre-coital interval (days) b 1.4 ± 0.7 2.3 ± 1.1 $4.2 \pm 3.3*$ Gestation length (days) 21.4 ± 0.6 21.9 ± 0.3 22.0 ± 0.5 Corpora lutea 15.6 ± 2.3 14.0 ± 1.4 15.1 ± 3.0 Implantation sites 15.0 ± 2.4 13.6 ± 1.1 13.0 ± 1.2 Number born 14.1 ± 1.9 12.9 ± 1.1 12.0 ± 1.6 Post-Implantation loss (%) d 6.0 5.1 7.7 a. Pregnant = uterine implantation sites. 5.1 7.7 b. Data summarized as mean \pm standard deviation. 5.1 7.7				900 mg/kg/day 12 11 11 11 3.8 ± 3.5 22.1 ± 0.5 13.8 ± 2.1 12.2 ± 3.7 10.8 ± 3.8° 11.5
		No manufaction officers and state of the sta	rified			
Results Remar	ks:	No reproductive effects were ident	inea.			

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Reliability:	1			
Reliability Remarks:	Reliable without restrictions.			
Key Study Sponsor Indicator:	Key study for the assessment of reproductive toxicity			
REFERENCE				
WIL Research (2012). WIL-402011. A Combined 28-Day Repeated Dose Oral (Gavage) Toxicity Study with the Reproduction/Developmental Toxicity Screening Test of Naphthenic Acid with a Mammalian Erythrocyte Micronucleus Rats. WIL Research Laboratories, LLC, Ashland, OH				

5. Toxicity Id Naphthenic Acids

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5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Species: RatSex: FemaleStrain: WistarRoute of admin.: Gavage

Exposure period

Frequency of treatm. : Daily

Duration of test

Doses : 0.6, 6 & 60 mg/kg/day

Control group

Method

Year : 2002 GLP : no data

Test substance : other TS: Naphthenic acid isolated from Athabasca oil sands tailings.

[Associated with CAS nymber 1338-24-5 in Toxline search]

Method : Oral doses of 60 mg/kg/day were given to female rats during

pre-breeding, breeding and gestation.

Result : The following description was given:

Reproductive toxicity testing demonstrated dramatic effects on female fertility at an oral dosage of 60 mg/kg/day during pre-breeding, breeding and gestation. While control and low dose (6 mg/kg/day) animals achieved 93 and 100% reproductive

success, respectively, only 7% of females dosed at 60 mg/kg/d successfully bore a litter. Total cholesterol of the latter group was 30% lower than controls. Mating and

ovulation were comparable amongst control and dose groups,

while fetal malformations were not apparent in any offspring. Results suggest that the dose-related infertility may be associated with poor embryonic implantation, an effect that might be secondary to depressed sex hormone

production requiring cholesterol as a precursor.

Reliability : (4) not assignable

This information is taken from an abstract. No other details of the study

could be obtained. The protocol of

the study does not appear to be comparable to a guideline

study, and the level of detail is insufficient to judge.

However, it may be useful in establishing dose levels for a

more in-depth study.

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DEVELOPMENTAL TOXICITY/TERATOGENICITY

TEST SUBSTANCE	
Category Chemical:	1338-24-5 Naphthenic acids
Test Substance:	1338-24-5 Naphthenic acids
Test Substance Purity/Composition and Other Test Substance Comments:	The test sample used in the current program was a blend of naphthenic acids from three sources. The samples were dried under a stream of nitrogen and then re-dissolved in 0.5 mL dichloromethane. The samples were analyzed by GC-MS (Young et al., 2008) and the total ion current mass spectra were collected and tablulated (Holowenko et al., 2002). Based on these data it was determined that there were no significant differences among these samples (Fedorak, 2009). The
	data indicated that the test material contained constituents with carbon numbers predominantly in the range of C6-C16 (corresponding to a molecular weight range of approximately 116-250) and with a ring distribution of approximately 0 rings (24%), 1 ring (39%), 2 rings (31%), 3 rings (5%) and 4 rings (1%).
Category Chemical Result Type :	Measured
Unable to Measure or Estimate Justification :	N/A
METHOD	
Route of Administration:	Oral
Other Route of Administration:	N/A
Type of Exposure:	Oral gavage
Species:	Rat
Other Species:	N/A
Mammalian Strain:	Sprague-Dawley
Other Strain:	N/A
Gender:	Female
Number of Animals per Dose:	12
Concentration:	The naphthenic acids were suspended in corn oil to the appropriate concentrations and administered in 10 ml/kg doses.
Dose:	100, 300, 900 mg/kg/day

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Year Study Performed :	2010
Method/Guideline Followed:	OPPTS 870.3650, 2000/OECD 422
GLP:	Yes. The testing was in accordance with Good Laboratory Practice Guidelines of the OECD (OECD, 1997) and the U.S. EPA (CFR, 2007).
Exposure Period:	Value or Lower Exposure Duration: 39 days Upper Exposure Duration: 53 days
	Dosing was initiated 14 days prior to mating and continued until post-natal day 3.
Frequency of Treatment:	Daily
Post-Exposure Period:	None
Method/Guideline and Test Condition Remarks:	The rats were obtained from Charles River Laboratories, Raleigh, North Carolina. They were held in the laboratory for a 16 day acclimation period and then randomly divided into treatment groups by weight. Mating was initiated after 14 days of dosing. Rats were mated on a 1:1 basis within each treatment group, females were co-housed with the males. Each mating pair was evaluated on a daily basis during the mating period. Successful mating was confirmed by the presence of a vaginal copulatory plug or the presence of sperm in a vaginal lavage. The day on which mating was confirmed was designated as gestational day 0. All females confirmed to have mated were placed in plastic maternity cages once mating was confirmed. Females for which copulation was not detected were placed in maternity cages at the end of the 14 day mating period. All females were allowed to deliver and to rear their young to post-natal day 4. On the day of parturition, all pups were examined for viability, for the presence of gross malformations and to assess gender. The numbers of live and stillborn pups were recorded. Length of gestation was calculated as the time from confirmation of mating to the onset of delivery. Females for which there was no evidence of mating were sacrificed on post-cohabitation day 25, those that showed evidence of mating but failed to deliver were euthanized on post-mating day 25, and all others were euthanized on post-natal day 4. Uteri with no microscopic evidence of implantation were opened and subsequently placed in 10% ammonium sulfide solution for detection of early implantation loss (Salewski, 1964).
	All offspring were uniquely identified and examined daily for signs of mortality and ill health. All offspring were individually weighed on post-natal days 1 and 4. Gender was assessed on post-natal days 0 and 4. At scheduled termination, post-natal day 4, all surviving offspring were euthanized and discarded without further examination. Parental mating, fertility, conception and copulation indices were analyzed using the Chi-square test with Yates' correction (Hollander and Wolfe, 1999). Mean parental body weights (weekly, gestation and lactation), body weight changes and food consumption, offspring body weights and body weight changes, gestation length, numbers of former implantation sites, numbers of corpora lutea, number of pups born, live litter size on PND 0, unaccounted for sites, and pre-coital intervals were evaluated by one-way analysis of variance (ANOVA) (Snedecor and Cochran, 1980) to determine intergroup differences between the vehicle control and test substance-treated groups. If the ANOVA revealed significant (p < 0.05) intergroup variance, Dunnett test (Dunnett, 1964) was used to compare the test substance-treated groups to the control group. Mean litter proportions (percent of litter) of males at birth and post-natal survival were evaluated using the Kruskal-Wallis

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nonparametric ANOVA (Kruskal and Wallis, 1952) to determine intergroup differences between the vehicle control and test substance-treated groups. If the ANOVA revealed significant (p < 0.05) intergroup variance, Dunn Test (Dunn, 1964) was used to compare the test substance-treated groups to the vehicle control group.

TEST RESULTS

	Concentration (LOAEL/LOAEC/NOAEL/NOAEC)					
Туре	Population:	Value Description:	Value or Lower Concentration:	Upper Concentration:		
	Units:	-				
NOAEL	Female Sprague-Daw	ley Rats	Maternal Effects 900	Mg/kg/day		
NOAEL	F1 offspring	Offspring delivered	300	Mg/kg/day		
NOAEL	F1 offspring	Offspring live born	100	Mg/kg/day		
NOAEL	F1 offspring	Offspring body weights	300	Mg/kg/day		

Results Remarks:

There were no apparent effects on mating. A single female in the 300 mg/kg/day group had a pre-coital interval of 13 days, resulting in a statistically significant increase in pre-coital incidence in this group. Otherwise all of the pairs productively mated and pre-coital intervals were within the historical control range for the laboratory. Note that there was a significant increase in pre-coital interval in the 300 mg/kg group, but this was due to a single female, and, as noted was within the historical range of the laboratory. Accordingly, it was not considered to have been a treatment-related effect.

The length of the gestational period was similar across the groups. There were reductions in the numbers of *corpora lutea* and implantation sites in the high dose group, but the differences were not statistically significant (see Table 1 below). However, there was a significant reduction in the number of offspring born/litter in the high dose group (Table 2). There was also a significant reduction in survival in offspring in the high dose group, and those that did survive had significantly lower body weights than the offspring in the control groups. The number of pups found dead or euthanized *in extremis* during the period PND 0-4 was: control = 1(1), 100 mg/kg/day = 0(0), 300 mg/kg/day = 12(5), and 900 mg/kg/day = 38(8).

Table 1. Summary of reproductive parameters assessed in the repeated dose/reproductive toxicity study of refined naphthenic acids.

Dose (mg/kg/day) Corn Oil Control	100 mg/kg/day	300 mg/kg/day	900 mg/kg/day
Number of females paired	12	12	12 12
Number of female mated	12	12	10 11
Number of females pregnant ^a	9	12	10 11
Number of females with litters	9	12	10 11
Pre-coital interval (days) ^b	1.4 + 0.7	2.3 + 1.1	4.2 + 3.3* 3.8 + 3.5
Gestation length (days)	21.4 <u>+</u> 0.6	21.9 <u>+</u> 0.3	22.0 <u>+</u> 0.5 22.1 <u>+</u> 0.5
Corpora lutea 15.6 ± 2.3	14.0 <u>+</u> 1.4	15.1 <u>+</u> 3.0	13.8 <u>+</u> 2.1
Implantation sites 15.0 ± 2.4	13.6 <u>+</u> 1.1	13.0 <u>+</u> 1.2	12.2 <u>+</u> 3.7
Number born 14.1 ± 1.9	12.9 <u>+</u> 1.1	12.0 <u>+</u> 1.6	10.8 <u>+</u> 3.8 ^c
Post-Implantation loss (%) ^d	6.0	5.1	7.7 11.5

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	 a. Pregnant = uterine implantation sites. b. Data summarized as mean + standard deviation. c. p < 0.05 Table 2. Survival, viability and growth of offspring following in utero exposure to refined naphthenic acids. The data are given as mean + SD.					
	Dose (mg/kg/day) Corn Oil 100 mg/kg/day 300 mg/kg/day 900 mg/kg/day					
	Number of viable litters 9 12 10 11 Number of pups born alive/litter 13.9 \pm 1.9 12.9 \pm 1.1 1 10.1 \pm 4.0 a 9.6 \pm 4.0 b Percentage of pups surviving from birth to termination 88.0 \pm 24.5 67.7 \pm 40.6 Pups (litters) found dead or euthanized in extremis 1(1) 0(0) 12(5) 38(8) Sex ratio (% males/litter) 58.9 \pm 9.6 53.9 \pm 9.6 55.2 \pm 19.1 58.1 \pm 22.7 Pup weight PND 1 - males 7.0 \pm 0.5 6.7 \pm 0.7 6.7 \pm 0.5 5.7 \pm 0.8 a Pup weight PND 4 - males 9.7 \pm 1.1 9.4 \pm 1.2 9.4 \pm 0.9 7.2 \pm 1.5 b Pup weight PND 4 - females 9.1 \pm 1.0 9.0 \pm 1.0 8.8 \pm 0.7 7.3 \pm 1.5 b a. P < 0.05, b. p < 0.01					
Conclusion:	Treatment of Sprague-Dawley rats with refined naphthenic acids had no apparent effects on mating and did not produce malformations at the highest dose tested (900 mg/kg/day). However, there were significant reductions in number of offspring, number live born and offspring body weights. The overall no observed adverse effect level was 100 mg/kg/day.					
RELIABILITY/DATA QUALITY						
Reliability:						
Reliability Remarks:	Reliable without restrictions.					
Key Study Sponsor Indicator:	Key study for the assessment of developmental toxicity					
REFERENCE						
Reference:	WIL Research (2012). WIL-402011. A Combined 28-Day Repeated Dose Oral (Gavage) Toxicity Study with the Reproduction/Developmental Toxicity Screening Test of Naphthenic Acid with a Mammalian Erythrocyte Micronucleus Test Rats. WIL Research Laboratories, LLC, Ashland, OH					

5. To	xicity		Naphthenic Acids May 15, 2012	
5.8.3	TOXICITY TO REPRODUCTION, OTHER	STUDIES		
5.9	SPECIFIC INVESTIGATIONS			
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6. Analyt. Meth. for Detection and Identification	Naphthenic Acids May 15, 2012
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7. E	ff. Against Target Org. and Intended Uses	Naphthenic Acids May 15, 2012
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7.1	FUNCTION	
7.2	EFFECTS ON ORGANISMS TO BE CONTROLLED	
7.3	ORGANISMS TO BE PROTECTED	
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8. Meas. Nec. to Prot. Man, Animals, Environment **Id** Naphthenic Acids **Date** May 15, 2012 8.1 METHODS HANDLING AND STORING 8.2 FIRE GUIDANCE **EMERGENCY MEASURES** 8.3 8.4 POSSIB. OF RENDERING SUBST. HARMLESS **WASTE MANAGEMENT** 8.6 SIDE-EFFECTS DETECTION 8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER 8.8 REACTIVITY TOWARDS CONTAINER MATERIAL

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9. References

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(1) AGS Chemicals Limited. (2003)
Material Safety Data Sheet, Naphthenic Acid.
Web Version URL: http://www.amtrade.co.uk/prodinfo.htm

(2) AGS Chemicals Limited. (2003).

Material Safety Data Sheet, Naphthenic Acid.

Web Version URL: http://www.amtrade.co.uk/prodinfo.htm

- (3) AGS Chemicals Limited. 2003. Material Safety Data Sheet, Naphthenic Acid. Web Version URL: http://www.amtrade.co.uk/prodinfo.htm
- Brient, J.A., P.J. Wessner, and M.N. Doyle. (1995)
 Naphthenic Acids.
 In: Kirk-Othmer Encyclopedia of Chemical Technology.
 John Wiley & Sons, Inc.
- (5) Brient, J.A., P.J. Wessner, and M.N. Doyle. 1995. Naphthenic Acids. In: Kirk-Othmer Encyclopedia of Chemical Technology.
 John Wiley & Sons, Inc.
- (6) Cairns, J. Jr., A. Scheier, and J.J. Loos. (1965)
 A comparison of the sensitivity to certain chemicals of adult zebra danios Brachydanio rerio (Hamilton-Buchanan) and zebra danio eggs with that of adult bluegill sunfish Lepomis macrochirus Raf.

 Notulae Naturae, No. 381:1-9.
- (7) Cairns, J. Jr., and A. Scheier. (1962)
 The effect of temperature and hardness of water upon the toxicity of naphthenic acids to the common bluegill (Lepomis macrochirus Raf.) and the pond snail (Physa heterostropha Say).
 Notulae Naturae. No. 353: 111 pp. Acad. Nat. Sci.

Philadelphia.

- (8) CEATAG (1998)
 Naphthenic acids background information discussion report.
 CEATAG (Conrad Environmental Aquatic Technical Advisory Group).
 Alberta Department of Energy, Edmonton, AB.
- (9) Dorn, P.B. (1992)
 Case Histories The petroleum refining industry.
 In: Ford, D.L. (ed.). Water Quality Management Library,
 Volume 3, Toxicity Reduction Evaluation and Control.
 Technomic Publishing Co., Lancaster, PA. pp 183 223.
- (10) Dourdoroff, P., B.G. Anderson, G.E. Burdick, P.S. Galstoff, W.B. Hart, T. Patrick, E.R. Strong, E.W. Surber, and W.M. VanHorn. (1951)
 Bioassay methods for the evaluation of acute toxicity of industrial wastes to fish.
 Sew. and Ind. Wastes. 23(11):1380-1397.
- (11) Exxon (1979)
 Acute Dermal Toxicity of MRD-79-10 in Rabbits
 MB 79-3702

9. References

Id Naphthenic Acids

Date May 15, 2012

(12) Exxon (1979)

Acute Oral Toxicity of MRD-79-10 in Rats,

MB 79-3702

(13) Exxon (1979)

Eye Irritation Study of MRD-79-10 in Rats

MB 79-3702

(14) Exxon Corporation. (1980)

Aquatic bioassay testing of Exxon Corporation's experimental

compounds (MRD 78-100).

Report by Battelle Columbus Laboratories, Columbus, Ohio.

(15) Harris, J.C. (1982)

Rate of hydrolysis.

In; Handbook of Chemical Property Estimation Methods. W.L. Lyman, W.F. Reehl, and D.H. Rosenblastt, eds.

Mcgraw-Hill Book Co., New York, NY.

(16) Hart, W.B., P. Doudoroff, and J. Greenbank. (1945)

The evaluation of the toxicity of the industrial wastes, chemicals and other substances to freshwater fishes

The Atlantic Refining Company, Philadelphia, PA. 315 pp.

(17) Havre, T.E. (2002)

Formation of calcium naphthenate in water/oil systems,

naphthenic acid chemistry and emulsion stability.

Ph.D. Thesis, Department of Chemical Engineering, Norwegian University of Science and Technology, Trondheim, Norway.

October 2002.

(18) Herman, D.C., P.M. Fedorak, and J.W. Costerton. (1993)

Biodegradation of cycloalkane carboxylic acids in oil sand

tailings.

Can. J. Microbiol. 39:576-580.

(19) Herman, D.C., P.M. Fedorak, M.D. MacKinnon, and J.W.

Costerton. (1994)

Biodegradation of naphthenic acids by microbial populations

indigenous to oils sands tailings.

(20) Hisaoka, K.K., and H.I. Battle. (1958)

The normal development stages of the zebra-fish, Brachydanio

rerio (Hamilton-Buchanan).

J. Morph. 102(2):311-327.

(21) Mackay, D. (1991)

Multimedia environmental models; The fugacity approach Lewis

Publ. CRC Press, Boca Raton, Florida

(22) Mallinckrodt Baker, Inc. (1997)

Material Safety Data Sheet No. N0310, Naphthenic Acids (CAS

No. 1338-24-5).

Mallinckrodt Baker Inc., Phillipsburg, New Jersey.

(23) Mallinckrodt Baker, Inc. (1997).

Material Safety Data Sheet No. N0310, Naphthenic Acids (CAS No. 1338-24-5).

Mallinckrodt Baker Inc., Phillipsburg, New Jersey.

(24) NTP. (2003)

http://ntp-server.niehs.nih.gov/htdocs/Overviews/GenProtocolsg.html.

9. References

Id Naphthenic Acids

Date May 15, 2012

(25) OECD (1995)

OECD Guideline 104, Vapor Pressure.

OECD, (Organization for Economic Cooperation and Development)

Paris, France.

(26) Peltier, W.H., and C.I. Weber, eds. (1985)

Method for measuring acute toxicity of effluents to freshwater and marine organisms, 3rd edition.

Environmental Monitoring and Support Laboratory, U.S. Environmental Protection Agency, Cincinnati, OH. EPA

600/4-85-014. 230 pp.

(27) Pennisi, S., and V.D. Lynch. (1977)

Pharmacologist 19: 181. [meeting abstract]

(28) Rockhold, W.T. (1955)

The toxicity of naphthenic acids and their metal salts.

Archs Ind Hlth 12, 477-482.

(30) Rogers, V.V., M. Wickstrom, K.Liber, and M.D. MacKinnon.

(2002)

Acute and subchronic mammalian toxicity of naphthenic acids

from oil sands tailings. Tox. Sci. 66: 347-355.

(31) Rogers, V.V., M. Wickstrom, K.Liber, and M.D. MacKinnon.

(2002)

Mammalian toxicity of naphthenic acids derived from the

Athabasca Oil Sands (AOS).

Toxicologist 66(1-S): 64-5. [meeting abstract]

(32) Shell Oil (1983)

Toxicity studies on oil additives: one generation reproduction study in male rabbits

repeatedly treated dermally with SAO Oil for 10 weeks

SBER.84.002

(33) SocTech, S.A. (2003)

Product Data Sheet, Naphthenic Acids.

Web Version URL:

http://www.soctech.ro/English/Produse/1acizinaft.htm

(34) SocTech, S.A. (2003).

Product Data Sheet, Naphthenic Acids.

Web Version URL: http://www.soctech.ro/English/Produse/1acizinaft.htm

(35) SocTech, S.A. 2003. Product Data Sheet, Naphthenic Acids. Web Version URL:

http://www.soctech.ro/English/Produse/1acizinaft.htm

(36) Stephan, C.E. (1977)

Method for calculating an LC50.

In: Aquatic Toxicology and Hazard Evaluation, ASTM STP 634.

American Society for Testing and Materials, Philadelphia,

PA. pp 65-84.

(37) U.S. EPA (2003)

Chemical Information Collection and Data Development

(Testing).

http://www.epa.gov/opptintr/chemtest/naphthst.htm.

ences		Naphthenic Acids May 15, 2012
U.S. EPA. (2000) EPI (Estimation Programs Interface) Suite, V 3.10, US Environmental Protection Agency, Office of pollution prevention and toxics, Washington DC.		
	EPI (Estimation Programs Interface) Suite, V 3.10, US Environmental Protection Agency, Office of pollution	U.S. EPA. (2000) EPI (Estimation Programs Interface) Suite, V 3.10, US Environmental Protection Agency, Office of pollution

10. Summary and Evaluation **Id** 1338-24-5 **Date** May 15, 2012 10.1 END POINT SUMMARY 10.2 HAZARD SUMMARY 10.3 RISK ASSESSMENT

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